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## (54) INTERMEDIATES FOR MACROCYCLIC COMPOUNDS

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This patent is subject to a terminal dis-

claimer.

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See application file for complete search history.

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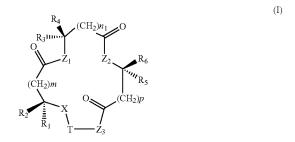
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#### (57) ABSTRACT

The present invention is directed to novel macrocyclic compounds of formula (I) and their pharmaceutically acceptable salts, hydrates or solvates:



wherein  $R_1$ ,  $R_2$ ,  $R_3$ ,  $R_4$ ,  $R_5$ ,  $R_6$ ,  $n_1$ , m, p  $Z_1$ ,  $Z_2$ , and  $Z_3$  are as describe in the specification. The invention also relates to compounds of formula (I) which are antagonists of the motilin receptor and are useful in the treatment of disorders associated with this receptor and with or with motility dysfunction.

#### 3 Claims, 21 Drawing Sheets

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General Synthetic Strategy to Conformationally-Defined Macrocycles of the Present Invention

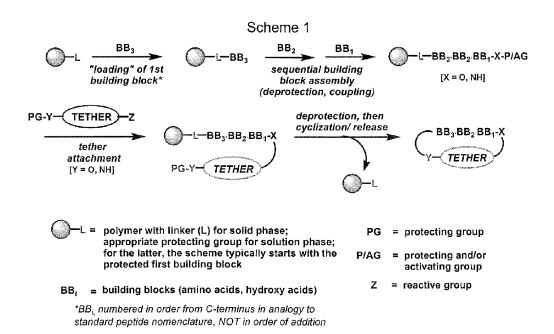


FIG. 1

FIG. 2

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The yield of **Ddz-T9** from **T9-0** on a 65 g scale was 60.9 g (91%)

<sup>1</sup>H NMR (CDCl<sub>3</sub>): δ 7.19-7.01, (m, 2H), 6.92-9.83 (m, 2H), 6.53 (bs, 2H), 6.34 (t, 1H), 5.17 (bt, 1H), 4.08 (m, 2H), 3.98 (m, 2H), 3.79 (s, 6H), 3.01 (bq, 2H), 2.66 (t, 3H), 1.26 (bs, 8H);

 $^{13}$ C NMR (CDCl<sub>3</sub>)  $\delta$  160.9, 156.8, 155.6, 149.6, 130.4, 127.5, 121.2, 111.7, 103.2, 98.4, 80., 69.7, 61.6, 55.5, 40.3, 30.5, 29.3, 27.4

FIG. 3

Standard Procedure for the Synthesis of Ddz-propargylamine

FIG. 4

# Standard Procedure for the Synthesis of Tether T10 *Method A*

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TLC (EtOAc/Hexanes 1:1, detection: UV, ninhydrin;  $R_f = 0.17$ )

<sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  7.18, t, 1H,J = 8.2Hz; 6.51, m, 5H; 6.34, t, 1H, J = 2.2Hz; 5.19, s, 1H; 4.05, t, 2H, J = 5.0Hz; 3.94, m, 4H; 3.75, s, 6H; 3.49, d, 2H J = 5.2Hz; 1.73, s, 6H.

<sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 160.856; δ 160.152; 160.005; 155.410; 149.305; 130.279; 107.438; 107.310; 103.163; 101.877; 98.517; 69.488; 67.382; 61.595; 55.427; 40.420; 29.427.

HPLC (standard gradient) t<sub>R</sub>: 7.25 min

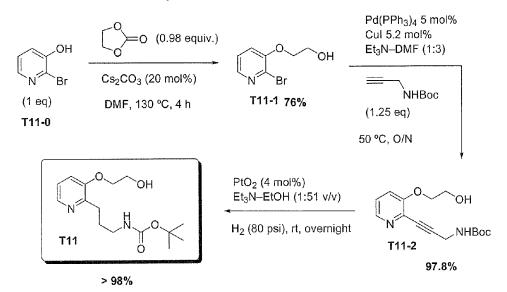
MS: 420 (M+H)

FIG. 5A

## Method B

The second synthetic route to **T10** is presented in the accompanying scheme.

FIG. 5B



TLC (15:85 THF/DCM; detection: UV; R<sub>f</sub>: 0.33).

<sup>1</sup>H NMR (DMSO-d<sub>6</sub>) δ 8.00, d, 1H; 7.32, d, 1H; 7.15, m, 1H; 6.44, s, 2H; 6.33, s, 1H; 3.99, t, 2H; 3.71, m, 8H; 2.89, m = 4, 2H; 2.71, t, 2H; 1.71, m = 5, 2H; 1.61, s, 6H.

 $^{13}$ C NMR, solvent DMSO-d<sub>6</sub>)  $\delta$  160.879; 153.275; 151.405; 150.447; 140.773; 122.666; 118.934; 103.347; 98.456; 79.778; 70.449; 60.212; 55.717; 55.599; 29.740; 28.592.

HPLC (standard gradient) t<sub>R</sub>: 5.4 min

MS: 419 (M+H)

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FIG. 7

Procedure for Synthesis of PPh<sub>3</sub>-DIAD Adduct

FIG. 8

## Standard Procedure for Attachment of Tethers via Reductive Amination

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FIG. 9

## Standard Procedure for the Synthesis of Tether T28

FIG. 10

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TLC (100% EtOAc; detection: UV, CMA;  $R_f$  = 0.24).

<sup>1</sup>H NMR (CDCl<sub>3</sub>, ppm): 7.74 (1H, dd), 7.35 (1H, d), 6.72 (1H, d), 6.53-6.49 (2H, m), 3.61-3.29 (1H, m), 5.06 (1H, t), 4.25-4.01 (2H, m), 3.91-3.89 (2H, m), 3.73 (3H, s), 2.99 (2H, dd), 2.63 (2H, t), 1.71 (8H, broad), 1.53 (9H, s).

<sup>13</sup>C NMR (CDCl<sub>3</sub>, ppm): 163.8, 162.2, 161.0, 159.7, 155.9, 149.4, 130.0, 129.1, 128.0, 126.8, 110.8, 98.1, 80.9, 79.3, 69.7, 61.3, 55.5, 39.1, 29.3, 28.5, 26.7.

FIG. 11

Standard Procedure for the Synthesis of Tether T33a and T33b

 $^{1}$ H NMR (CDCl<sub>3</sub>) δ (ppm) 7.18-7.11 (m, 2H), 6.90 (m, 2H), 6.52 (m, 2H), 6.33(m, 1H), 5.09 (bt, 1H), 4.52 (m, 1H), 3.77 (s, 6H), 3.08 (bq, 2H), 2.64 (bt, 2H), 1.75 (m, 8H); 1.27 (bd, 3H),  $^{13}$ C NMR (CDCl<sub>3</sub>) δ 160.8, 155.5, 149.5, 131.2, 130.6, 127.4, 121.2, 113.3, 103.2, 98,4, 80.7, 74.8, 66.5, 55,4, 40.2, 30.6, 29.3, 29.2, 27.4, 16.1

FIG. 12A

FIG. 12B

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TLC (100% EtOAc; detection: CMA,  $R_f = 0.5$ ).

MW Calc. for  $C_{24}H_{35}N_3O_7$ , 477.55; MS Found  $\left(M+H\right)^{+}$  478.

<sup>1</sup>H NMR (CDCl<sub>3</sub>) § 1.62 (m, 2H), 1.70 (m, 8H), 2.43 (m, 2H), 2.67 (m, 2H), 3.07 (m, 2H), 3.34 (s, 3H), 3.43 (s, 3H), 3.61 (m, 2H), 3.75 (s, 6H), 5.40 (sb, 1H), 6.31 (s, 1H), 6.49 (s, 2H)

<sup>13</sup>C NMR (CDCl<sub>3</sub>) &523.25 ( $\&Cemtharpoonup{C}{CH}_2$ ), 28.56 ( $\&Cemtharpoonup{C}{CH}_3$ ), 39.31 ( $\&Cemtharpoonup{C}{CH}_3$ ), 30.09 ( $\&Cemtharpoonup{C}{CH}_3$ ), 31.25 ( $\&Cemtharpoonup{C}{CH}_2$ ), 40.16 ( $\&Cemtharpoonup{C}{CH}_3$ ), 61.38 ( $\&Cemtharpoonup{C}_3$ ), 61.38 (&Cemt

HPLC (standard gradient) t<sub>R</sub>: 6.68 min.

FIG. 13

TLC (25/75 EtOAc/Hex; detection: UV, ninhydrin;  $R_f = 0.03$ )

 $^{1}$ H NMR (CDCl<sub>3</sub>):  $\delta$  7.06-7.00 (bt, 1H), 6.61-6.52 (m, 4H), 6.35 (m, 1H), 5.12 (bt, 1H), 4.03 (m, 2H), 3.95 (m, 2H), 3.77 (s, 6H), 3.11-3.04 (bq, 2H), 2.60 (bt, 2H), 1.75 (m, 8H)

<sup>13</sup>C NMR (CDCl<sub>3</sub>): δ 163.9, 160.9, 160.6, 157.6, 157.5, 155.6, 149.5, 130.8, 130.6, 125.9, 107.26, 106.9, 103.2, 98,4, 80.8, 77.5, 69.9, 61,3, 60.9, 60.6, 55,4, 40.3, 30.4, 29.3, 26.9,

HPLC (standard gradient): t<sub>R</sub> = 8.37 min

FIG. 14

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### 74.5% (2 steps)

TLC: (25/75 EtOAc/Hex; detection: UV, ninhydrin;  $R_f = 0.03$ )

 $^{1}$ H NMR (CDCl<sub>3</sub>)  $\delta$  (ppm): 6.84-6.75 (m, 3H), 6.52 (bs, 2H), 6.34 (m, 1H), 5.17 (bt, 1H), 4.01 (m, 2H), 3.93 (m, 2H), 3.77 (s, 6H), 3.10 (bq, 2H), 2.63 (bt, 2H), 1.74 (m, 8H)

<sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 160.9, 158.9, 155.8, 155.6, 152.9, 152.9, 149.5, 132.4, 132.3, 117.1, 116.8, 112.7, 112.6, 103.2, 98.4, 80.8, 70.4, 61.6, 55.5, 40.2, 30.3, 29.3, 27.4.

HPLC (standard gradient): t<sub>R</sub> = 8.29 min

FIG. 15

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TLC (25/75 EtOAc/Hex; detection: UV, ninhydrin;  $R_f = 0.03$ )

 $^{1}$ H NMR (CDCl<sub>3</sub>):  $\delta$  7.12-7.08 (bd, 2H), 6.76-6.73 (d, 1H), 6.52 (m, 2H), 6.33 (bs, 1H), 5.15 (bt, 1H), 4.02 (m, 2H), 3.95 (m, 2H), 3.79 (s, 6H), 3.09 (bq, 2H), 2.61 (bt, 2H), 1.74 (m, 8H).  $^{13}$ C NMR (CDCl $_3$ )  $\delta$  160.8, 155.6, 155.4, 149.5, 132.4, 130.1, 127.0, 126.0, 112.8, 103.2, 98.4, 80.8, 70.0, 61.4, 55.5, 40.3, 30.2, 29.3, 24.5, 27.4

HPLC (standard gradient): t<sub>R</sub> = 9.60 min

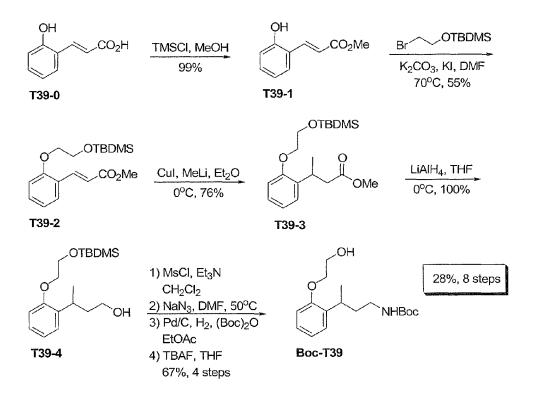
FIG. 16

 $^{1}$ H NMR (CDCl<sub>3</sub>):  $\delta$  7.20-7.10, (m,  $^{2}$ H), 6.95-6.80 (m,  $^{2}$ H), 6.55 (bs,  $^{2}$ H), 6.35 (s,  $^{1}$ H), 5.18 (bt,  $^{1}$ H), 4.12 (m, 1H), 3.95 (m, 2H), 3.80 (s, 6H), 3.15 (bq, 2H), 2.65 (t, 2H), 1.98 (bs, 2H), 1.65 (bs, 6H), 1.25 (m, 3H).

<sup>13</sup>C NMR (CDCl<sub>3</sub>): §160.8, 156.6, 155.8, 149.6, 130.4, 127.5, 121.3, 111.7, 103.2, 98.4, 80.7, 73.5, 66.6, 55.5, 40.2, 30.5, 29.3, 29.1, 27.3, 19.5.

FIG. 17

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TLC (50% EtOAc, 50% Hex; detection: UV and CMA;  $R_f = 0.25$ )

<sup>1</sup>H NMR (CDCl<sub>3</sub>, ppm): 7.11-7.08 (2H, m), 6.86 (1H, t), 6.76 (1H, d), 5.05 (1H, broad), 4.26-3.85 (4H, m), 3.22-3.07 (2H, m), 2.71 (1H, broad), 1.66-1.60 (2H, m), 1.33 (9H, s), 1.17 (3H, d).

<sup>13</sup>C NMR (CDCl<sub>3</sub>, ppm): 156.1, 135.0, 127.1, 127.0, 121.4, 111.7, 69.9, 61.5, 39.8, 38.4, 28.7, 20.7.

FIG. 18

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TLC (50% EtOAc, 50% Hex; detection: UV and CMA; R<sub>f</sub> = 0.25)

<sup>1</sup>H NMR (CDCl<sub>3</sub>, ppm): 7.11-7.08 (2H, m), 6.86 (1H, t), 6.76 (1H, d), 5.05 (1H, broad), 4.26-3.85 (4H, m), 3.22-3.07 (2H, m), 2.71 (1H, broad), 1.66-1.60 (2H, m), 1.33 (9H, s), 1.17 (3H, d).

FIG. 19

<sup>&</sup>lt;sup>13</sup>C NMR (CDCl<sub>3</sub>, ppm): 156.1, 135.0, 127.1, 127.0, 121.4, 111.7, 69.9, 61.5, 39.8, 38.4, 28.7, 20.7.

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TLC (100% EtOAc; detection: CMA;  $R_f = 0.5$ )

<sup>1</sup>H NMR (CDCl<sub>3</sub>) δ.1.23 (s, 3H), 1.49 (s, 3H), 1.69 (s, 3H), 1.74 (s, 3H), 1.90 (m, 2H), 2.35 (m, 1H), 3.35 (m, 2H), 3.76 (s, 6H), 3.92 (m, 2H), 4.40 (m, 2H), 5.10 (m, 1H), 6.15 (s, 1H), 6.25 (s, 2H).

<sup>13</sup>C NMR (CDCI<sub>3</sub>) &25.52 ( $\&Cine{C}$ H<sub>3</sub>), 27.53 ( $\&Cine{C}$ H<sub>3</sub>), 28.88 ( $\&Cine{C}$ H<sub>3</sub>), 29.61 ( $\&Cine{C}$ H<sub>3</sub>), 35.92 ( $\&Cine{C}$ H<sub>2</sub>), 42.62 ( $\&Cine{C}$ H<sub>2</sub>), 55.43  $(\underline{CH}_3), \ 60.60 \ (\underline{CH}_2), \ 82.38 \ (\underline{CH}), \ 83.33 \ (\underline{CH}), \ 83.68 \ (\underline{CH}), \ 84.96 \ (\underline{CH}), \ 98.26 \ (\underline{CH}), \ 103.23 \ (\underline{CH}), \ 118.3$ (Cq), 149.50 (Cq), 156.20 (Cq), 160, 02 (Cq)

HPLC (standard gradient): t<sub>R</sub> = 6.64 min

MS: M+H found: 439

FIG. 20

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 $^{1}$ H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  6.82-6.98 (m, 2H); 6.80-6.75 (m, 1H); 6.53 (s, 2H); 6.35 (t, 1H, 2 Hz); 5.23 (b, 1H); 4.08 (m, 1H); 3.90-3.68 (m, 8H); 3.20-2.97 (m, 2H); 2.95-53 (m, 4H); 2.0-1.63 (m, 10H).

<sup>13</sup>C NMR (75.5 MHz, CDCl<sub>3</sub>) δ160.85; 155.56; 152.55; 149.56; 128.13; 127.77; 120.28; 103.22; 98.43; 80.72; 76.80; 65.76; 55.46; 40.23; 30.45; 29.34; 29.22; 27.10; 24.97; 23.94.

Scheme 2: Thioester Strategy for Macrocycl'ic Compounds of the Present Invention

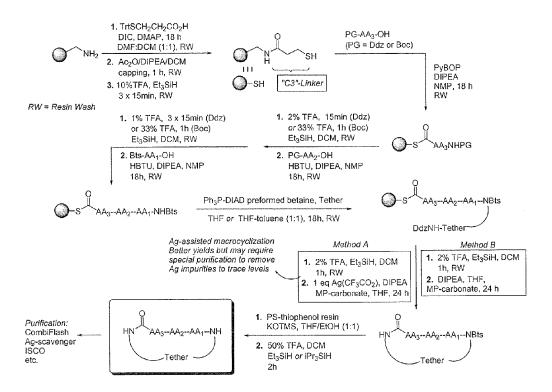


FIG. 22

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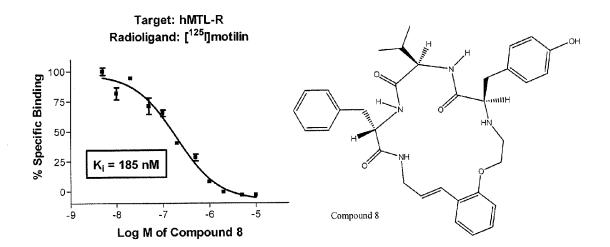


FIG. 23

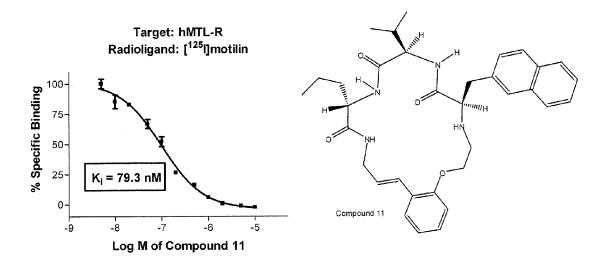


FIG. 24

### INTERMEDIATES FOR MACROCYCLIC **COMPOUNDS**

#### RELATED APPLICATION INFORMATION

This application is a continuation under 35 U.S.C. §120 of U.S. patent application Ser. No. 10/872,142, filed Jun. 18, 2004, now U.S. Pat. No. 7,521,420, which claims the benefit of U.S. Patent Application Ser. No. 60/479,223, filed Jun. 18, 2003. The disclosure of each application is incorporated 10 receptor (WO 99/64436) has simplified and accelerated the herein by reference in its entirety.

#### FIELD OF THE INVENTION

The present invention relates to novel conformationallydefined macrocyclic compounds, pharmaceutical compositions comprising same and intermediates used in their manufacture. More particularly, the invention relates to macrocyclic compounds that have been demonstrated to invention further relates to macrocyclic compounds useful as therapeutics for a range of gastrointestinal disorders, in particular those in which malfunction of gastric motility or increased motilin secretion is observed, such as hypermotilinemia, irritable bowel syndrome and dyspepsia.

#### BACKGROUND OF THE INVENTION

A number of peptide hormones are involved in the control of the different functions in the gastrointestinal (GI) tract, 30 including absorption, secretion, blood flow and motility (Mulvihill, et al. in Basic and Clinical Endocrinology, 4th edition, Greenspan, F. S.; Baxter, J. D., eds., Appleton & Lange: Norwalk, Conn., 1994, pp 551-570). Since interactions between the brain and GI system are critical to the 35 proper modulation of these functions, these peptides can be produced locally in the GI tract or distally in the CNS.

One of these peptide hormones, motilin, a linear 22-amino acid peptide, plays a critical regulatory role in the GI physimotor activity. As such, the peptide is periodically released from the duodenal mucosa during fasting in mammals, including humans. More precisely, motilin exerts a powerful effect on gastric motility through the contraction of gastrointestinal smooth muscle to stimulate gastric emptying, 45 decrease intestinal transit time and initiate phase III of the migrating motor complex in the small bowel (Itoh, Z., Ed., Motilin, Academic Press: San Diego, Calif., 1990, ASIN: 0123757304; Nelson, D. K. Dig. Dis. Sci. 1996, 41, 2006-2015; Peeters, T. L.; Vantrappen, G.; Janssens, J. Gastroen- 50 terology 1980, 79, 716-719).

Motilin exerts these effects through receptors located predominantly on the human antrum and proximal duodenum, although its receptors are found in other regions of the GI tract as well (Peeters, T. L.; Bormans, V.; Vantrappen, G. Regul. 55 Pept. 1988, 23, 171-182). Therefore, motilin hormone is involved in motility of both the upper and lower parts of the GI system (Williams et al. Am. J. Physiol. 1992, 262, G50-G55). In addition, motilin and its receptors have been found in the CNS and periphery, suggesting a physiological role in the 60 nervous system that has not yet been definitively elucidated (Depoortere, I.; Peeters, T. L. Am. J. Physiol. 1997, 272, G994-999 and O'Donohue, T. L et al. Peptides 1981, 2, 467-477). For example, motilin receptors in the brain have been suggested to play a regulatory role in a number of CNS 65 functions, including feeding and drinking behavior, micturition reflex, central and brain stem neuronal modulation and

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pituitary hormone secretion (Itoh, Z. Motilin and Clinical Applications. Peptides 1997, 18, 593-608; Asakawa, A.; Inui, A.; Momose, K.; et al., M. Peptides 1998, 19, 987-990 and Rosenfeld, D. J.; Garthwaite, T. L. Physiol. Behav. 1987, 39, 753-756). Physiological studies have provided confirmatory evidence that motilin can indeed have an effect on feeding behavior (Rosenfeld, D. J.; Garthwaite, T. L. Phys. Behav. 1987, 39, 735-736)

The recent identification and cloning of the human motilin search for agents which can modulate its activity for specific therapeutic purposes.

Due to the critical and direct involvement of motilin in control of gastric motility, agents that either diminish (hypomotility) or enhance (hypermotility) the activity at the motilin receptor, are a particularly attractive area for further investigation in the search for new effective pharmaceuticals towards these indications.

Peptidic agonists of the motilin receptor, which have cliniselectively antagonize the activity of the motilin receptor. The 20 cal application for the treatment of hypomotility disorders, have been reported (U.S. Pat. Nos. 5,695,952; 5,721,353; 6,018,037; 6,380,158; 6,420,521, U.S. Appl. 2001/0041791, WO 98/42840; WO 01/00830 and WO 02/059141). Derivatives of erythromycin, commonly referred to as motilides, 25 have also been reported as agonists of the motilin receptor (U.S. Pat. Nos. 4,920,102; 5,008,249; 5,175,150; 5,418,224; 5,470,961; 5,523,401, 5,554,605; 5,658,888; 5,854,407; 5,912,235; 6,100,239; 6,165,985; 6,403,775).

Antagonists of the motilin receptor are potentially extremely useful as therapeutic treatments for diseases associated with hypermotility and hypermotilinemia, including irritable bowel syndrome, dyspepsia, gastroesophogeal reflux disorders, Crohn's disease, ulcerative colitis, pancreatitis, infantile hypertrophic pyloric stenosis, diabetes mellitus, obesity, malabsorption syndrome, carcinoid syndrome, diarrhea, atrophic colitis or gastritis, gastrointestinal dumping syndrome, postgastroenterectomy syndrome, gastric stasis and eating disorders leading to obesity.

A variety of peptidic compounds have been described as ological system though governing of fasting gastrointestinal 40 antagonists of the motilin receptor (Depoortere, I.; Macielag, M. J.; Galdes, A.; Peeters, T. L. Eur. J. Pharmacol. 1995, 286, 241-247; U.S. Pat. Nos. 5,470,830; 6,255,285; 6,586,630; 6,720,433; U.S. 2003/0176643; WO 02/64623). These peptidic antagonists suffer from the known limitations of peptides as drug molecules, in particular poor oral bioavailability and degradative metabolism.

> Cyclization of peptidic derivatives is a method employed to improve the properties of a linear peptide both with respect to metabolic stability and conformational freedom. Cyclic molecules tend to be more resistant to metabolic enzymes. Such cyclic tetrapeptide motilin antagonists have been reported (Haramura, M. et al J. Med. Chem. 2002, 45, 670-675, U.S. 2003/0191053; WO 02/16404).

> Other motilin antagonists, which are non-peptidic and noncyclic in nature have also been reported (U.S. Pat. Nos. 5,972, 939; 6,384,031; 6,392,040; 6,423,714; 6,511,980; 6,624,165; 6,667,309; U.S. 2002/0111484; 2001/041701; 2002/ 0103238; 2001/0056106, 2002/0013352; 2003/0203906 and 2002/0002192)

> The macrocyclic motilin antagonists of the present invention comprise elements of both peptidic and non-peptidic structures in a combination which has not been pursued for this application previously.

> Indeed, the structural features of antagonists of the present invention are different. In particular, within the known motilin antagonists which are cyclic peptides, it was found that such derivatives containing D-amino acids were devoid of

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activity. In contrast, for the tripeptidomimetic compounds of the present invention, the D-stereochemistry is required for two of the three building elements.

The motilin antagonists of the present invention are also distinct from the prior art in that they comprise a tether element to fulfill the dual role of controlling conformations and providing additional sites for interaction either through hydrophobic interactions, hydrogen bonding or dipole-dipole interactions.

#### SUMMARY OF THE INVENTION

In a first aspect, the present invention is directed to compounds of formula (I):

$$\begin{array}{c} R_4 \\ R_2 \\ R_2 \\ R_2 \\ R_1 \end{array} \begin{array}{c} (CH_2)n_1 \\ Z_2 \\ R_6 \\ (CH_2)p \\ R_5 \end{array} \begin{array}{c} (I) \\ Z_0 \\ R_6 \\ (CH_2)p \\ Z_1 \end{array} \begin{array}{c} (CH_2)n_1 \\ Z_2 \\ (CH_2)p \\ Z_2 \end{array} \begin{array}{c} (I) \\ Z_2 \\ (I) \\ Z_2 \\ (I) \\ Z_3 \\ (I) \\ Z_4 \\ (I) \\ Z_5 \\ (I) \\ Z_6 \\ (I) \\ Z_7 \\ (I) \\ Z_8 \\ (I) \\ Z_9 \\ (I) \\ (I) \\ Z_9 \\ (I) \\ (I)$$

and pharmaceutically acceptable salts, hydrates or solvates thereof wherein:

 $Z_1$ ,  $Z_2$  and  $Z_3$  are independently selected from the group consisting of O, N and NR<sub>10</sub>, wherein R<sub>10</sub> is selected from the group consisting of hydrogen, lower alkyl, and substituted lower alkyl;

 $R_{\rm 1}$  is independently selected from the group consisting of lower alkyl substituted with aryl, lower alkyl substituted with substituted with heteroaryl and lower alkyl substituted with substituted heteroaryl;

R<sub>2</sub> is hydrogen;

 $R_3$  is independently selected from the group consisting of alkyl and cycloalkyl with the proviso that when  $Z_1$  is  $N,\,R_3$  can form a four, five, six or seven-membered heterocyclic ring together with  $Z_1$ ;

R<sub>4</sub> is hydrogen;

 $\rm R_5$  and  $\rm R_6$  are independently selected from the group consisting of hydrogen, alkyl, substituted alkyl, cycloalkyl, substituted cycloalkyl, heterocyclic, substituted heterocyclic, aryl, substituted aryl, heteroaryl and substituted heteroaryl, with the proviso that at least one of  $\rm R_5$  and  $\rm R_6$  is hydrogen;

X is selected from the group consisting of O,  $NR_8$ , and  $N(R_9)_2^+$ ;

wherein  $R_8$  is selected from the group consisting of hydrogen, lower alkyl, substituted lower alkyl, formyl, acyl, carboxyalkyl, carboxylaryl, amido, sulfonyl, sulfonamido and amidino; and

R<sub>9</sub> is selected from the group consisting of hydrogen, lower slkyl, and substituted lower alkyl;

m,  $n_1$  and p are independently selected from 0, 1 or 2; and T is a bivalent radical of formula II:

$$-U-(CH_2)_d-W-Y-Z-(CH_2)_e-$$
 (II) 60

wherein d and e are independently selected from 0, 1, 2, 3, 4 or 5;

wherein U is bonded to X of formula (I) and is —CH<sub>2</sub>— or —C(—O)—;

wherein Y and Z are each optionally present;

W, Y and Z are independently selected from the group consisting of: —O—, —NR<sub>28</sub>—, —S—, —SO—,

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$$G_1$$
 $G_1$ 
 $G_2$ 
 $G_1$ 
 $G_2$ 
 $G_1$ 
 $G_2$ 
 $G_1$ 
 $G_2$ 
 $G_1$ 
 $G_2$ 
 $G_1$ 
 $G_2$ 
 $G_2$ 
 $G_1$ 
 $G_2$ 
 $G_2$ 
 $G_3$ 
 $G_4$ 
 $G_5$ 
 $G_5$ 
 $G_7$ 
 $G_8$ 
 $G_9$ 
 $G_9$ 

wherein any carbon atom contained within said ring structure, can be replaced by a nitrogen atom, with the proviso that if said ring structure is a monocyclic ring structure, it does not comprise more than four nitrogen atoms and if said ring structure is a bicyclic ring structure, it does not comprise more than six nitrogen atoms;

 $G_1$  and  $G_2$  each independently represent a covalent bond or a bivalent radical selected from the group consisting of -O-,  $-NR_{41}-$ , -S-, -SO-,  $-SO_2-$ , -C(=O)-, -C(=O)-, -C(=O)-, -C(=O)-, -C(=O)-, -C(=O)-,  $-SO_2-$ NH-,  $-NH-SO_2-$ ,  $-CR_{42}R_{43}-$ , -CH=CH- with a configuration Z or E, and -C=C-; with the proviso that  $G_1$  is bonded closer to U than  $G_2$ ;

K<sub>1</sub>, K<sub>2</sub>, K<sub>3</sub>, K<sub>4</sub>, K<sub>6</sub>, K<sub>15</sub> and K<sub>16</sub> are independently selected from the group consisting of O, NR<sub>44</sub> and S;

f is selected from 1, 2, 3, 4, 5 or 6;

R<sub>31</sub>, R<sub>32</sub>, R<sub>38</sub>, R<sub>39</sub>, R<sub>48</sub> and R<sub>49</sub> are independently selected from hydrogen, halogen, alkyl, substituted alkyl, cycloalkyl, substituted cycloalkyl, heterocyclic, substituted heterocyclic, aryl, substituted aryl, heteroaryl, substituted heteroaryl, hydroxy, alkoxy, aryloxy, amino, halogen, formyl, acyl, carboxy, carboxyalkyl, carboxyaryl, amido, carbamoyl, guanidino, ureido, amidino, cyano, nitro, mercapto, sulfinyl, sulfonyl and sulfonamido; and

R<sub>33</sub>, R<sub>34</sub>, R<sub>35</sub>, R<sub>36</sub>, R<sub>37</sub>, R<sub>47</sub>, R<sub>50</sub> and R<sub>51</sub> are independently selected from hydrogen, halogen, alkyl, substituted alkyl, cycloalkyl, substituted cycloalkyl, heterocyclic, substituted heterocyclic, aryl, substituted aryl, heteroaryl, substituted heteroaryl, hydroxy, alkoxy, aryloxy, oxo, amino, halogen, formyl, acyl, carboxy, carboxyalkyl, carboxyaryl, amido, carbamoyl, guanidino, ureido, amidino, cyano, nitro, mercapto, sulfinyl, sulfonyl and sulfonamido.

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In a second aspect, the invention also proposes compounds of formula (I) which are antagonists of the motilin receptor.

In a third aspect, the invention proposes a method of treating a disorder associated with the motilin receptor or motility dysfunction in humans and other mammals, comprising administering a therapeutically effective amount of a compound of formula (I).

While the invention will be described in conjunction with example embodiments, it will be understood that it is not intended to limit the scope of the invention to such embodiment. On the contrary, it is intended to cover all alternatives, modifications and equivalents as may be included as defined by the appended claims.

#### DETAILED DESCRIPTION OF THE INVENTION

Preferably in formula (I), as depicted hereinabove,  $R_1$  is selected from the group consisting of  $-(CH_2)_qR_{11}$ , and  $-CHR_{12}R_{13}$ 

wherein q is 0, 1, 2 or 3; and

R<sub>11</sub> and R<sub>12</sub> are independently selected from a ring structure from the following group:

F  $A_1$   $A_2$   $A_3$   $A_5$   $A_5$   $A_5$   $A_5$   $A_5$   $A_5$   $A_5$   $A_7$   $A_8$   $A_$ 

wherein any carbon atom in said ring structure can be replaced a nitrogen atom, with the proviso that if said ring structure is a monocyclic ring structure, it does not comprise more than four nitrogen atoms and if said ring 65 structure is a bicyclic ring structure, it does not comprise more than six nitrogen atoms;

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A<sub>1</sub>, A<sub>2</sub>, A<sub>3</sub>, A<sub>4</sub> and A<sub>5</sub> are each optionally present and are independently selected from the group consisting of halogen, alkyl, substituted alkyl, cycloalkyl, substituted cycloalkyl, heterocyclic, substituted heterocyclic, aryl, substituted aryl, heteroaryl, substituted heteroaryl, hydroxy, alkoxy, aryloxy, amino, halogen, formyl, acyl, carboxy, carboxyalkyl, carboxyaryl, amido, carbamoyl, guanidino, ureido, amidino, cyano, nitro, mercapto, sulfinyl, sulfonyl and sulfonamido;

B<sub>1</sub>, B<sub>2</sub>, B<sub>3</sub>, and B<sub>4</sub> are independently selected from NR<sub>14</sub>, S or O, wherein R<sub>14</sub> is selected from the group consisting of hydrogen, alkyl, substituted alkyl, formyl, acyl, carboxyalkyl, carboxyaryl, amido, sulfonyl and sulfonamido:

 $R_{13}$  is as defined for as  $R_{11}$  and  $R_{12}$  or is selected from the group comprising lower alkyl, substituted lower alkyl, hydroxy, alkoxy, aryloxy, amino, carboxy, carboxyalkyl, carboxyaryl, and amido.

20 wherein A<sub>1</sub>, A<sub>2</sub>, A<sub>3</sub>, A<sub>4</sub> and A<sub>5</sub> are most preferably selected from halogen, trifluororomethyl, C<sub>1-6</sub> alkyl or C<sub>1-6</sub> alkoxy.

Preferably,  $R_{11}$ ,  $R_{12}$  and  $R_{13}$  are selected from the group consisting of:

wherein  $R_a$  and  $R_b$  are chosen from the group consisting of Cl, F, CF<sub>3</sub>, OCH<sub>3</sub>, OH, and C(CH<sub>3</sub>)<sub>3</sub> and CH<sub>3</sub>.

Also preferably, R<sub>3</sub> in formula (I), is selected from the group consisting of:

—(CH<sub>2</sub>)<sub>s</sub>CH<sub>3</sub>, —CH(CH<sub>3</sub>)(CH<sub>2</sub>)<sub>t</sub>CH<sub>3</sub>, —CH(OR<sub>15</sub>)CH<sub>3</sub>, —CH<sub>2</sub>SCH<sub>3</sub>—CH<sub>2</sub>CH<sub>2</sub>SCH<sub>3</sub>, —CH<sub>2</sub>S(=O)CH<sub>3</sub>,

 $-\text{CH}_2\text{CH}_2\text{S}(=0)\text{CH}_3,$   $-\text{CH}_2\text{S}(=0)_2\text{CH}_3,$ 

-CH<sub>2</sub>CH<sub>2</sub>S(=O)<sub>2</sub>CH<sub>3</sub>, -(CH<sub>2</sub>)<sub>u</sub>CH(CH<sub>3</sub>)<sub>2</sub>,

 $-C(CH_3)_3$ , and  $-(CH_2)_y$ ,  $-R_{21}$ , wherein: s and u are independently selected from 0, 1, 2, 3, 4 or 5;

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t is independently selected from 1, 2, 3 or 4; y is selected from 0, 1, 2, 3 or 4;

 $R_{15}$  is selected from the group consisting of hydrogen, alkyl, substituted alkyl, formyl and acyl;

R<sub>21</sub> is selected from a ring structure selected from the following group:

$$E_1$$
 $E_2$ 
 $E_3$ 
 $E_2$ 
 $E_3$ 
 $E_4$ 
 $E_5$ 
 $E_5$ 
 $E_5$ 
 $E_7$ 
 $E_8$ 
 $E_9$ 
 $E_9$ 

wherein any carbon atom in said ring structure can be replaced by a nitrogen atom, with the proviso that if said ring structure is a monocyclic ring structure, it does not comprise more than four nitrogen atoms and if said ring structure is a bicyclic ring structure, it does not comprise more than six nitrogen atoms;

z is selected from 1, 2, 3, 4 or 5;

E<sub>1</sub>, E<sub>2</sub> and E<sub>3</sub> are each optionally present and are independently selected from the group consisting of halogen, alkyl, substituted alkyl, cycloalkyl, substituted cycloalkyl, heterocyclic, substituted heterocyclic, aryl, substituted aryl, heteroaryl, substituted heteroaryl, hydroxy, alkoxy, aryloxy, amino, halogen, formyl, acyl, carboxy, carboxyalkyl, carboxyaryl, amido, carbamoyl, guanidino, ureido, amidino, cyano, nitro, mercapto, sulfinyl, sulfonyl and sulfonamido; and

J is optionally present and is selected from the group consisting of alkyl, substituted alkyl, cycloalkyl, substituted cycloalkyl, heterocyclic, substituted heterocyclic, aryl, substituted aryl, heteroaryl, substituted heteroaryl, hydroxy, alkoxy, aryloxy, oxo, amino, halogen, formyl, acyl, carboxy, carboxyalkyl, carboxyaryl, amido, carbamoyl, guanidino, ureido, amidino, mercapto, sulfinyl, sulfonyl and sulfonamido.

The tether portion (T) of formula (I) is preferably selected from the group consisting of:

$$(X) \qquad \qquad L_1 \qquad \qquad (Z_3) \qquad \qquad (Z_3) \qquad \qquad (Z_3) \qquad \qquad (X) \qquad$$

8

-continued 
$$(X)$$
  $(Z_3)$   $(X)$   $(Z_3)$   $(X)$   $(Z_3)$   $(X)$   $(Z_3)$   $(X)$   $(Z_3)$   $(Z_3)$   $(Z_3)$ 

wherein  $L_1$  is O, NH or NMe;  $L_2$  is CH or N;  $L_3$  is CH or N;  $L_4$  is O or CH<sub>2</sub>;  $L_5$  is CH or N  $L_6$  is CR<sub>52</sub>R<sub>53</sub> or O; R<sub>46</sub> is H or CH<sub>3</sub>:

 $R_{52}$ ,  $R_{53}$ ,  $R_{54}$ ,  $R_{55}$ ,  $R_{56}$  and  $R_{57}$  are independently selected from hydrogen, lower alkyl, substituted lower alkyl, hydroxy, alkoxy, aryloxy, amino, and oxo; or  $R_{52}$  together with  $R_{53}$  or  $R_{54}$  together with  $R_{57}$  can independently form a three to seven-membered cyclic ring comprising carbon, oxygen, sulfur and/or nitrogen atoms;

(X) is the site of a covalent bond to X in formula (I); and ( $Z_3$ ) is the site of a covalent bond to  $Z_3$  in formula (I).

In a particularly preferred embodiment of the invention, there are provided compounds of formula (I) wherein m, n and p are 0, X,  $Z_1$ ,  $Z_2$  and  $Z_3$  are NH and  $R_2$ ,  $R_4$  and  $R_5$  are hydrogen, represented by formula (III):

According to another aspect of the invention, there are provided compounds of formula (I) wherein when  $Z_1$  is a nitrogen atom,  $R_3$  forms a four, five, six or seven-membered heterocyclic ring together with  $Z_1$ , represented by formula (IV):

wherein said heterocyclic ring may contain a second nitrogen atom, or an oxygen, or sulfur atom;

 $n_2$  is selected from 0, 1, 2 or 3

 $R_{7}$  is optionally present and is selected from the group consisting of alkyl, substituted alkyl, cycloalkyl, substituted cycloalkyl, heterocyclic, substituted heterocyclic, aryl, substituted aryl, heteroaryl, substituted heteroaryl, hydroxy, 20 alkoxy, aryloxy, oxo, amino, halogen, formyl, acyl, carboxy, carboxyalkyl, carboxyaryl, amido, carbamoyl, guanidino, ureido, amidino, mercapto, sulfinyl, sulfonyl and sulfonamido.

It is to be understood, that in the context of the present <sup>25</sup> invention, the terms amino, guanidine, ureido and amidino encompass substituted derivatives thereof as well.

Preferably, the invention provides a method of treating a disorder associated with hypermotility or hypermotilinemia in humans and other mammals comprising administering a therapeutically effective amount of a compound of formula (I).

#### BRIEF DESCRIPTION OF THE DRAWINGS

FIG. 1 depicts Scheme 1 presenting a general synthetic strategy to conformationally-defined macrocycles of the present invention.

FIG. 2 depicts the standard procedure for the synthesis of  $_{\rm 40}$  tether T8 of Example 16.

FIG. 3 depicts the standard procedure for the synthesis of tether T9 of Example 17.

FIG. 4 depicts the standard procedure for the synthesis of Ddz-propargylamine of Example 18.

FIG. 5A depicts the standard procedure for the synthesis of tether T10 of Example 19.

FIG. **5**B depicts the second synthetic route to tether T10 of Example 19.

FIG. 6 depicts the standard procedure for the synthesis of 50 Tether T11 of Example 20.

FIG. 7 depicts the standard procedure for the synthesis of tether T12 of Example 26.

FIG. 8 depicts the procedure for synthesis of PPh<sub>3</sub>-DIAD adduct of Example 29-C.

FIG. 9 depicts the standard procedure for attachment of tethers via reductive amination of Example 30.

FIG. 10 depicts the standard procedure for the synthesis of tether T28 of Example 32.

FIG. 11 the standard procedure for the synthesis of tether 60 T32 of Example 36.

FIGS. 12A, 12B depict the standard procedure for the synthesis of tether T33a and T33b of Example 37.

FIG. 13 depicts the standard procedure for the synthesis of tether T34 of Example 38.

FIG. **14** depicts the standard procedure for the synthesis of tether T35 of Example 39.

FIG. 15 depicts the standard procedure for the synthesis of tether T36 of Example 40.

FIG. 16 depicts the standard procedure for the synthesis of tether T37 of Example 41.

FIG. 17 depicts the standard procedure for the synthesis of tether T38 of Example 42.

Chiral T38 can be accessed through the use of asymmetric synthesis methods, resolution or chiral chromatography techniques available in the literature.

HPLC (standard gradient)  $t_R$ =8.46 min

Chiral material can be accessed by starting with the chiral epoxide. For example, the (S)-isomer of T38 was constructed in 89% overall yield from (S)-propylene oxide.

FIG. 18 depicts the standard procedure for the synthesis of tether T39 of Example 43. Chiral T39 can be accessed through the use of asymmetric synthesis methods, resolution or chiral chromatography techniques available in the literature

FIG. 19 depicts the standard procedure for the synthesis of tether T40 of Example 44.

Chiral T40 can be accessed through the use of asymmetric synthesis methods, resolution or chiral chromatography techniques available in the literature.

FIG. **20** depicts the standard procedure for the synthesis of tether T41 of Example 45.

FIG. 21 depicts the standard procedure for the synthesis of tether T42 of Example 46.

FIG. **22** depicts Scheme 2 of the thioester strategy for macrocyclic compounds of the present invention.

FIG. 23 depicts the competitive binding curve for compound 8.

FIG. **24** depicts the competitive binding curve for compound 11

## DESCRIPTION OF PREFERRED EMBODIMENTS

Although preferred embodiments of the present invention have been described in detail herein and illustrated in the accompanying structures, schemes and tables, it is to be understood that the invention is not limited to these precise embodiments and that various changes and modifications may be effected therein without departing from the scope or spirit of the present invention.

Specifically preferred compounds of the present invention, include, but are not limited to:

-continued

-continued

In addition to the preferred tethers (T) illustrated previously, other specific tethers employed for compounds of the invention are shown hereinbelow:

45

T35

T28

T33a [(R)-isomer] T33b [(S)-isomer]

-continued

PG and PG' indicate a standard amine protecting group compatible with synthetic protocol, such as Boc, Ddz, Fmoc, or Alloc

In a preferred embodiment, the present invention is directed to a method of treating irritable bowel syndrome, dyspepsia, Crohn's disease, gastroesophogeal reflux disorders, ulcerative colitis, pancreatitis, infantile hypertrophic pyloric stenosis, carcinoid syndrome, malabsorption syndrome, diarrhea, diabetes mellitus, obesity, postgastroenterectomy syndrome, atrophic colitis or gastritis, gastric stasis, gastrointestinal dumping syndrome, celiac disease and eating disorders leading to obesity in humans and other mammals comprising administering a therapeutically effective amount of a compound of formula (I).

#### Synthetic Methods

A. General Information

Reagents and solvents were of reagent quality or better and were used as obtained from various commercial suppliers 50 unless otherwise noted. DMF, DCM and THF used are of DriSolv® (EM Science, now EMD Chemicals, Inc., part of Merck KgaA, Darmstadt, Germany) or synthesis grade quality except for (i) deprotection, (ii) resin capping reactions and (iii) washing. NMP used for the amino acid (AA) coupling 55 reactions is of analytical grade. DMF was adequately degassed by placing under vacuum for a minimum of 30 min prior to use. Tyr(3tBu) was synthesized following the method reported in JP2000 44595. Cpa was made using literature methods (Tetrahedron: Asymmetry 2003, 14, 3575-3580) or 60 obtained commercially. Boc- and Fmoc-protected amino acids and side chain protected derivatives, including those of N-methyl and unnatural amino acids, were obtained from commercial suppliers or synthesized through standard methodologies known to those in the art. Ddz-amino acids were 65 either synthesized by standard procedures or obtained commercially from Orpegen (Heidelberg, Germany) or Advanced ChemTech (Louisville, Ky., USA). Bts-amino acids were

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synthesized as described in Example 6. Hydroxy acids were obtained from commercial suppliers or synthesized from the corresponding amino acids by literature methods. Analytical TLC was performed on pre-coated plates of silica gel 60F254 (0.25 mm thickness) containing a fluorescent indicator. The 5 term "concentrated/evaporated under reduced pressure" indicates evaporation utilizing a rotary evaporator under either water aspirator pressure or the stronger vacuum provided by a mechanical oil vacuum pump as appropriate for the solvent being removed. "Dry pack" indicates chromatography on silica gel that has not been pre-treated with solvent, generally applied on larger scales for purifications where a large difference in R<sub>e</sub> exists between the desired product and any impurities. For solid phase chemistry processes, "dried in the standard manner" is that the resin is dried first in air (1 h), and subsequently under vacuum (oil pump usually) until full dryness is attained ( $\sim$ 30 min to O/N).

# B. Synthetic Methods for Building Blocks of the Invention

#### Example 6

Standard Procedure for the Synthesis of Bts-Amino Acids

$$R_{AA}$$
 $R_{AA}$ 
 $R_{AA}$ 
 $R_{AA}$ 
 $R_{AA}$ 
 $R_{AA}$ 
 $R_{AA}$ 
 $R_{AA}$ 
 $R_{AA}$ 
 $R_{AA}$ 

To a solution of the amino acid or amino acid derivative (0.1 mol, 1.0 eq) in 0.25 N sodium hydroxide (0.08 mol, 0.8 eq) with an initial pH of approximately 9.5 (pH meter) at rt, solid Bts-Cl (0.11 mol, 1.1 eq) was added in one portion. The resulting suspension was stirred vigorously for 2-3 d. The pH of the reaction should be adjusted with 5.0 N sodium hydroxide as required to remain within the range 9.5-10.0 during this time. Typically, the pH has to be adjusted every 20-30 min during the first 5 h. Once the pH stops dropping, it is an indication that the reaction is almost complete. This can be 50 confirmed by TLC (EtOAc:MeOH, 95:5). Upon completion, the reaction mixture was washed with Et<sub>2</sub>O. Washing is continued until the absence of non-polar impurities in the aqueous layer is confirmed by TLC (typically 3×100 mL). The aqueous solution was then cooled to 0° C., acidified to pH 2.0 55 Step T8-1: Chlorotrimethylsilane (116 mL, 0.91 mol, 1.5 eq) with 1 N HCl until no additional cloudiness forms, and extracted with EtOAc (3×100 mL). Alternatively, a mixture of DCM and EtOAc may be used as the extraction solvent, depending on the solubility of the product obtained from different amino acids or derivatives. Note that DCM cannot 60 be used solely as solvent because of the emulsion formed during extraction. The combined organic phases were washed with brine (2×150 mL), dried over MgSO<sub>4</sub>, filtered and evaporated under reduced pressure. DCM (1x) and hexanes (2x) were evaporated from the residue in order to ensure complete removal of the EtOAc and give the desired compound as a solid in 55-98% yield.

The following are modifications that have proven useful for certain amino acids: Gly, Ala, D-Ala, β-Ala and GABA: Use 1.5 eq of amino acid per eq of Bts-Cl, in order to prevent dibetsvlation.

Met: Carry out the reaction under N<sub>2</sub> to prevent oxidation. Gln and Asn: Due to the solubility of Bts-Gln and Bts-Asn, the work-up required is modified from the standard procedure: Upon completion of the reaction, the reaction mixture was washed with diethyl ether. Washing is continued until the absence of non-polar impurities in the aqueous layer is confirmed by TLC (typically 3×100 mL). The aqueous phase was then cooled to 0° C. and acidified to pH 2.0 with 6 N HCl. 6 N HCl was employed to minimize the volume of the solution due to the water solubility of Bts-Gln and Bts-Asn. (They are, in contrast, difficult to dissolve in DCM, EtOAc or chloroform.) The solution was maintained at 0° C. for 10 min and the product was collected by filtration as a white precipitate. The solid was washed with cold water (1x), cold brine (2x) and water (1×, 25° C.). The pH of this wash was taken, if it is not <sup>20</sup> approximately 4, the solid was washed again with water. Finally, the solid was washed with cold EtOAc, then with cold Et<sub>2</sub>O (2×), and finally dried under vacuum (oil pump) (83-85% yield).

### C. General Synthetic Strategy to Conformationally-Defined Macrocycles of the Present Invention

The compounds of Formula I can be synthesized using 30 traditional solution synthesis techniques or solid phase chemistry methods. In either, the construction involves four phases: first, synthesis of the building blocks, including one to four moieties, comprising recognition elements for the biological target receptor, plus one tether moiety, primarily for control and definition of conformation. These building blocks are assembled together, typically in a sequential fashion, in a second phase employing standard chemical transformations. The precursors from the assembly are then cyclized in the third stage to provide the macrocyclic structures. Finally, a post-cyclization processing stage involving removal of protecting groups and optional purification then provides the desired final compounds (see FIG. 1). This method has been previously disclosed in WO 01/25257 and U.S. patent application Ser. No. 09/679,331. A general synthetic strategy is shown in FIG. 1.

### D. Procedures for the Synthesis of Representative Tethers of the Present Invention

The important tether component required for compounds of the invention are synthesized as described in WO01/25257, U.S. Provisional Pat. Appl. Ser. No. 60/491,248 or herein. A standard procedure for the synthesis of tether B is shown in FIG. 2.

was added to a suspension of 2-hydroxycinnamic acid (100 g, 0.61 mol, 1.0 eq) in MeOH (500 mL, HPLC grade) over 30 min at 0° C. The resulting mixture was stirred at rt O/N. The reaction was monitored by TLC (EtOAc/MeOH: 98/2). Heating the reaction mixture in a hot water can accelerate the process if necessary. After the reaction was completed, the reaction mixture was evaporated under reduced pressure to afford methyl 2-hydroxycinnamate as a white solid (108.5 g) in quantitative yield. The identity of this intermediate compound is confirmed by NMR. This reaction can be carried out on larger (kg) scale with similar results

Step T8-2: 3,4-Dihydro-2H-pyran (DHP, 140 mL, 1.54 mol, 2.52 eq) was added dropwise to 2-bromoethanol (108 mL, 1.51 mol, 2.5 eq) in a 2 L three-neck flask with mechanical stirring at 0° C. over 2 h. The resulting mixture was stirred for additional 1 h at rt. Methyl 2-hydroxycinnamate from 5 Step T8-1 (108 g, 0.61 mol, 1.0 eq), potassium carbonate (92.2 g, 0.67 mol, 1.1 eq), potassium iodide (20 g, 0.12 mol, 0.2 eq) and DMF (300 mL, spectrometric grade) were added to the above flask. The reaction mixture was stirred at 70° C. (external temperature) for 24 h. The reaction was monitored by TLC (DCM/Et<sub>2</sub>O: 95/5). The reaction was allowed to cool to rt and Et<sub>2</sub>O (450 mL) was added. The inorganic salts were removed by filtration and washed with  $Et_2O(3\times50 \,\mathrm{mL})$ . The filtrate was diluted with hexanes (400 mL) and washed with water (3×500 mL), dried over 15 MgSO<sub>4</sub>, filtered and the filtrate evaporated under reduced pressure. The crude ester (desired product and excess Br—C<sub>2</sub>H<sub>4</sub>—OTHP) was used for the subsequent reduction without further purification.

Step T8-3: DIBAL (1.525 L, 1.525 mol, 2.5 eq, 1.0 M in 20 DCM) was added slowly to a solution of the above crude ester from Step T8-2 (0.61 mol based on the theoretical yield) in anhydrous DCM (610 mL) at -35° C. with mechanical stirring over 1.5 h. The resulting mixture was stirred for 1.5 h at  $-35^{\circ}$  C., then 1.5 h at  $0^{\circ}$  C. The reaction 25 was monitored by TLC (hex/EtOAc:50/50). When complete, Na<sub>2</sub>SO<sub>4</sub>.10H<sub>2</sub>O (100 g, 0.5 eq) was slowly added; hydrogen evolution was observed, when it subsided water was added (100 mL). The mixture was warmed to rt and stirred for 10 min, then warmed to 40° C. with hot water 30 and stirred under reflux for 20 min. The mixture was cooled to rt, diluted with DCM (600 mL), and the upper solution decanted into a filter. The solid that remained in the flask was washed with dichloromethane (5×500 mL) with mechanical stirring and filtered. The filtrate from each 35 wash was checked by TLC, and additional washes performed if necessary to recover additional product. In an alternative work-up procedure, after dilution with DCM (600 mL), the mixture was filtered. The resulting solid was then continuously extracted with 0.5% TEA in dichlo- 40 romethane using a Soxhlet extractor. Higher yield was typically obtained by this alternative procedure, although it does require more time. The filtrate was concentrated under reduced pressure and the residue purified by dry pack (EtOAc/hex/Et<sub>3</sub>N: 20/80/0.5) to give the product 45 alcohol as a yellowish oil (yield: 90%). The identity and purity were confirmed by NMR.

Step T8-4: To a mixture of the allylic alcohol from Step T8-3 (28 g, 0.100 mol, 1.0 eq) and collidine (0.110 mol, 1.1 eq) in 200 mL of anhydrous DMF under N<sub>2</sub> was added anhy- 50 drous LiCl (4.26 g, 0.100 mol, 1.0 eq.) dissolved in 100 mL of anhydrous DMF. The mixture was then cooled to 0° C., and MsCl (12.67 g, 0.110 mol, 1.1 eq., freshly distilled over P<sub>2</sub>O<sub>5</sub>), was added dropwise. The reaction was allowed to warm to rt and monitored by TLC (3:7 EtOAc/hex). When 55 the reaction was complete, NaN<sub>3</sub> (32.7 g, 0.500 mol, 5.0 eq.) was added. The reaction mixture was stirred at rt O/N with progress followed by NMR. When the reaction was complete, the mixture is poured into an ice-cooled water bath, and extracted with diethyl ether (3x). The combined 60 organic phases were then washed sequentially with citrate buffer  $(2\times)$ , saturated sodium bicarbonate  $(2\times)$ , and finally with brine  $(1\times)$ . The organic layer was dried with MgSO<sub>4</sub>, filtered and the filtrate concentrated under reduced pressure. The allylic azide was obtained in 90% combined 65 yield, and was of sufficient quality to use as such for the following step.

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Step T8-5:  $PPh_3$  (25.9 g, 0.099 mol, 1.5 eq) was added at  $0^{\circ}$  C. to a solution of the allylic azide from Step T8-4 (20.0 g, 0.066 mol, 1.0 eq.) in 100 mL of THF. The solution was stirred for 30 min at 0° C. and 20 h at rt. Water (12 mL) was then added and the resulting solution was heated at 60° C. for 4 h. The solution was cooled to rt, 2N HCl (15 mL) added and the mixture stirred for 90 min at 50° C. The separated organic phase was extracted with 0.05 N HCl (2×100 mL). The combined aqueous phase was washed with Et<sub>2</sub>O (5×150 mL) and toluene (4×150 mL) (more extraction could be necessary, follow by TLC), which were combined and back-extracted with 0.05 N HCl (1×100 mL). This acidic aqueous phase from back-extraction was combined with the main aqueous phase and washed with ether (5×150 mL) again. The pH of the aqueous phase was then adjusted to 8-9 by the addition of sodium hydroxide (5 N). Care must be exercised to not adjust the pH above 9 due to the reaction conditions required by the next step. The aqueous phase was concentrated under reduced pressure (aspirator, then oil pump) or lyophilized to dryness. Toluene  $(2\times)$  was added to the residue and then also evaporated under reduced pressure to remove traces of water. The crude product (desired amino alcohol along with inorganic salt) was used for the next reaction without further purifi-

Step T8-6: A mixture of the crude amino alcohol from Step T8-5 (0.5 mol based on the theoretical yield), Ddz-OPh (174 g, 0.55 mol, 1.1 eq) and Et<sub>3</sub>N (70 mL, 0.5 mol, 1.0 eq) in DMF (180 mL) was stirred for 24 h at 50° C. Additional DMF is added if required to solubilize all materials. The reaction was monitored by TLC (hex/EtOAc:50/50, ninhydrin detection). After the reaction was complete, the reaction mixture was diluted with Et<sub>2</sub>O (1.5 L) and water (300 mL). The separated aqueous phase was extracted with Et<sub>2</sub>O (2×150 mL). The combined organic phase was washed with water ( $3\times500 \text{ mL}$ ) and brine ( $1\times500 \text{ mL}$ ), dried over MgSO<sub>4</sub>, filtered and the filtrate concentrated under reduced pressure. The layers were monitored by TLC to ensure no product was lost into the aqueous layer. If so indicated, perform one or more additional extractions with Et<sub>2</sub>O of the aqueous phase to recover this material. The crude product was purified by dry pack (recommended column conditions: EtOAc/hex/Et<sub>3</sub>N: 35/65/0.5 to 65/35/ 0.5) to give the tether Ddz-T8 as a pale yellow syrup (yield: ~40%). The identity and purity of the product was confirmed by NMR.

<sup>1</sup>H NMR (DMSO-d<sub>6</sub>): 1.6 ppm (s, 6H, 2×CH3), 3.6-3.8 ppm (wide s, 10H, 2×OCH<sub>3</sub>, 2×OCH<sub>2</sub>), 3.95 ppm (triplet, 2H, CH<sub>2</sub>N), 6-6.2 ppm (m, 2H, 2×CH), 6.2-6.5 ppm (m, 3H, 3×CH, aromatic), 6.6-7.6 ppm (m, 5H, aromatic).

A standard procedure for the synthesis of tether T9 is shown in FIG. 3.

Tether T9 can also be synthesized from T8 by reduction as in step T9-3 or with other appropriate hydrogenation catalysts known to those in the art.

A standard procedure for the synthesis of Ddz propargylamine is shown in FIG. 4,

In a dried three-neck flask, a solution of propargylamine (53.7 g, 0.975 mol, 1.5 eq) in degassed DMF (Drisolv, 388 mL) was treated with Ddz-N<sub>3</sub> (170.9 g, 0.65 mol, 1.0 eq), tetramethylguanidine (TMG, 81.4 mL, 0.65 mol, 1.0 eq) and DIPEA (113.1 mL, 0.65 mol, 1.0 eq) and stirred at 50° C. O/N. The reaction was monitored by TLC (conditions: 25/75 EtOAc/hex. R<sub>j</sub>: 0.25; detection: UV, ninhydrin). Upon completion, DMF was evaporated under reduced pressure until dryness and the residue dissolved in Et<sub>2</sub>O (1 L). The organic solution was washed sequentially with citrate buffer

(pH 4.5, 3×), saturated aqueous sodium bicarbonate (2×), and brine (2×), then dried with MgSO<sub>4</sub>, filtered and the filtrate evaporated under reduced pressure. A pale orange solid was obtained. This solid was triturated with 1% EtOAc in hex, then collected by filtration and dried under vacuum (oil 5 pump) to provide the desired product (153.4 g, 85.2%).

A standard procedure for the synthesis of tether T10 is shown in FIG. 5A.

Two alternative routes to this tether have been developed. The first synthetic approach proceeded starting from the commercially available monobenzoate of resorcinol (T10-0). Mitsunobu reaction under standard conditions with the protected amino alcohol from Example 9, followed by saponification of the benzoate provided T10-1 in good yield after recrystallization. Alkylation of the phenol with 2-bromoethanol using the optimized conditions shown permitted the desired product Ddz-T10 to be obtained after dry pack purification in 42% yield.

A second synthetic route to T10 is shown in FIG. 5B.

From resorcinol, two successive Mitsunobu reactions are 20 conducted with the appropriate two carbon synthons illustrated, themselves derived from 2-aminoethanol and ethylene glycol, respectively, through known protection methodologies. Lastly, deprotection of the silyl ether, also under standard conditions provided Boc-T10.

Although the yields in the two methods are comparable, the first required less mechanical manipulation and is preferred for larger scales.

A standard procedure for the synthesis of tether T11 is shown in FIG. 6.

A standard procedure for the synthesis of tether T12 is shown in FIG. 7.

In a 3-L flame-dried three-neck flask, a solution of (aminomethyl)phenylthiobenzyl alcohol (12-0, 96 g, 0.39 mol) in degassed DMF (1 L, 0.4 M) was prepared. To this was added 35 DdzN<sub>3</sub> (0.95 eq), followed by TMG (0.39 mol, 49 mL). The reaction was stirred for 10 min, then DIPEA (68 mL, 0.39 mol) added. The mixture was heated at 50° C. under N<sub>2</sub> until TLC indicated no DdzN<sub>3</sub> remained (48 h typically). (TLC eluent: EtOAc:Hex 50:50; detection: ninhydrin). Upon 40 completion, to the reaction mixture was added 3 L citrate buffer and the separated aqueous layer extracted with Et<sub>2</sub>O (3×1500 mL). The combined organic phase was washed sequentially with citrate buffer (2×200 mL), water (2×200 mL) and brine (2×200 mL). The organic layer was dried over 45 MgSO<sub>4</sub>, filtered and the filtrate evaporated under reduced pressure. A dark orange oil was obtained, which was purified by dry-pack. For this procedure, the oil was first dissolved in EtOAc:Hex:DCM:TEA (20:80:1:0.5, v/v/v/v). At this point, a little extra DCM was sometimes required to ensure com- 50 plete dissolution. The solution was loaded onto the column, then the column eluted with EtOAc:Hex:DCM:Et<sub>3</sub>N (20:80: 1:0.5) until all the impurities were separated out as indicated by TLC, paying particular attention to that closest to the desired product. The elution was then continued with EtOAc: 55 Hex:Et<sub>3</sub>N 30:70:0.5 (v/v/v) and finally with EtOAc:hexanes: Et<sub>3</sub>N (50:50:0.5) to elute the desired product. After removal of the solvent from the fractions containing the product under reduced pressure, the residue was dissolved in the minimum amount of DCM, a three-fold larger volume of hexanes added, then the solvents again evaporated under reduced pressure. This treatment was repeated until an off-white foam was obtained. The latter solidified while drying under vacuum (oil pump). Alternatively, the material yielded a solid after sequential concentration with DCM  $(1\times)$  and hexanes  $(2\times)$ . 65 Tether Ddz-T12 was obtained as an off-white solid (85-90% yield).

# Example 29

Standard Procedure for Attachment of Tethers Utilizing the Mitsunobu Reaction

#### Example 29-A

# Using PPh<sub>3</sub>-DIAD Isolated Adduct

To a 0.2 M solution of the appropriate tether (1.5 eq) in THF or THF-toluene (1:1) was added the PPh<sub>3</sub>-DIAD (preformed by mixing equivalent amounts of the reagents and isolated by evaporation of solvent, see Example 29-C) adduct (1.0 eq.). The resultant mixture was manually agitated for 10 sec (the solution remained turbid), then added to the resin. Alternatively, the resin was added to the solution. The reaction suspension was agitated O/N (after ~5 min the mixture becomes limpid). The resin was filtered and washed 2×DCM, 1× toluene, 1× EtOH, 1× toluene, 1× (DCM/MeOH), 1× (THF/MeOH), 2×DCM, then dried in the standard manner.

## Example 29-B

# Using "PPh3-DIAD In Situ Procedure"

To a 0.2 M solution of the appropriate tether (4 eq) in THF or THF-toluene (1:1) was added triphenylphosphine (4 eq). The resultant mixture was manually shaken until a homogenous solution was obtained, then added to the resin. Alternatively, the resin (or IRORI<sup>TM</sup> MiniKans® (NEXUS Biosystems, Poway, Calif.), miniaturized microreactors, containing resin) was added to the solution. To this suspension was then added DIAD (3.9 eq) and the reaction agitated O/N. Note: Since the reaction is exothermic, for larger scales, the reaction should be cooled in an ice bath. In addition, an appropriate vent must be supplied to allow any pressure build-up to be released. The resin was filtered and washed DCM (2×), toluene (1×), EtOH (1×), toluene (1×), DCM/MeOH (1×), 1×THF/MeOH (1×), DCM/MeOH (1×), 2×DCM, then dried in the standard manner.

A procedure for the synthesis of PPh<sub>3</sub>-DIAD adduct is shown in FIG. 8.

DIAD (1 eq) was added dropwise to a well-stirred solution of triphenylphosphine (1 eq) in THF (0.4 M) at 0° C. under nitrogen. The mixture was then maintained at 0° C. with stirring for 30 min. The white solid obtained was collected by filtration (use medium sized fritted filters), washed with cold anhydrous THF until the washes were colorless, and lastly washed once with anhydrous  $\rm Et_2O$ . The white solid product was then vacuum-dried (oil pump) and stored under nitrogen. (Note: The PPh<sub>3</sub>-DIAD adduct can be made in larger than immediately required quantity and stored under nitrogen; it is very important to store this reagent under anhydrous conditions.)

# Example 30

#### Standard Procedure for Attachment of Tethers via Reductive Amination as shown in FIG. 9

In certain instances, the Mitsunobu process of Example 29 cannot be applied or is not efficient for incorporation of the tether. Hence, reductive amination has been developed as an alternative that can be employed for tether incorporation as illustrated hereinbelow for one of the preferred tethers. Similar chemistry can be used to incorporate other tethers of the present invention.

34 Example 47

Standard Procedure for Macrocyclization with Thioester Linker

The resin containing the cyclization precursor is combined in an appropriate vessel with pre-washed MP-carbonate resin [Argonaut Technologies, Foster City, Calif., commercially supplied MP-carbonate resin was treated with 3×THF (1 L per 400 g) and dried O/N at 30° C. in a vacuum oven] (1.4 to 1.6 eq relative to the initial loading of the synthesis resin). A 0.2 M DIPEA solution in THF was then added to the combined resins (1 mL/60 mg MP-carbonate resin) and the suspension agitated O/N at rt. Subsequently, the resin was filtered and rinsed 2×THF. The combined filtrates are collected together in an appropriate vessel, then the volatile contents evaporated under reduced pressure [in addition to the standard methods, solvent can also be removed in vacuo using centrifugal evaporation (ThermoSavant Discovery®, Speed-Vac® or comparable) (Thermo Electron Corporation, 20 Waltham, Mass.)] to provide the crude macrocycles.

#### Example 48

Standard Procedure for Silver-Assisted Macrocyclization with Thioester Linker

Except for the cyclization itself and subsequent work-up, this procedure is identical to that of Example 47. The resin containing the cyclization precursor was combined in an appropriate vessel with pre-washed MP-carbonate resin [Argonaut Technologies, commercially supplied MP-carbonate resin was treated with THF (3×, 1 L per 400 g) and dried O/N at 30° C. in a vacuum oven] (1.4 to 1.6 eq relative to the initial loading of the synthesis resin). To this was added THF (1 mL per 100 mg resin) and silver trifluoroacetate (1 eq relative to the initial loading of the resin). Finally, an amount of DIPEA sufficient to obtain a 0.2 M solution was added. The reaction mixture was agitated at rt O/N. The solution was then filtered and the resins washed 2×THF. The filtrates are collected together in an appropriate vessel, then evaporated under reduced pressure [(the volatile contents could also be removed in vacuo using centrifugal evaporation (ThermoSavant Discovery®, SpeedVace or comparable)] to provide the crude macrocycles. For this procedure, silver trifluoroacetate should be stored in a dessicator between uses. In addition, it is recommended to use a new bottle of THF (or a bottle that has been recently opened under N<sub>2</sub> or Ar) to minimize formation of silver oxide.

Additionally, a ring-closing metathesis (RCM) strategy, as developed by Grubbs et al. can also be used to access some of the macrocyclic compounds of the invention (see for example U.S. Pat. No. 5,811,515; Grubbs, R. H. et al. *J. Org. Chem.* 2001, 66, 5291-5300; Furstner, A. Angew. Chem. Int. Ed. 2000, 39, 3012-3043).

To access certain derivatives of compounds of the present invention, additional reactions from those in the general scheme were required. For some, it was advantageous to react the functionality to be derivatized prior to the formation of the macrocyclic ring. The cyclic structure can restrict access of reagents to that functionality. For example, in the synthesis of N-methyl and N-acyl derivatives of macrocycles, where the secondary nitrogen atom of the ring is the site of derivatization, the reaction is preferred to be performed prior to the application of the appropriate cyclization protocol.

In other cases, for example the derivatization of side chain functionality, the reaction was best performed after formation of the macrocyclic ring. For example, further reaction of amino moieties on side chains examples was typically efficiently done by reaction of the partially protected macro-

The Tether (30-2) with the amine protected as its Ddz derivative was efficiently oxidized to the corresponding aldehyde 30-2 using SO<sub>3</sub>-pyr in DMSO-Et<sub>3</sub>N-DCM. This aldehyde (0.14 mmol, 56 mg, 1.5 eq based upon loading of resin support) was dissolved in a 1:3 mixture of TMOF-MeOH (DriSolv, 4 mL) at rt. To this was added the resin containing the tripeptide (30-1, as its trifluoroacetic acid salt from the deprotection of the terminal amine), the mixture was agitated briefly to wet the resin, and then borane-pyridine complex (as the commercially available 8 M solution, 23 µL, 2 eq) was introduced to the suspension. The reaction was agitated O/N, then the resin filtered, washed with DCM  $(2\times)$ , THF  $(1\times)$ , DCM/MeOH [3:1] (1x), THF/MeOH [3:1] (1x), DCM (2x) and dried in the standard manner. Care must be taken to ensure that the desired resin bound product 30-3 is not contaminated with the dialkylated material. However, even if the reaction does not proceed to completion or if a small amount of the dialkylation side product is present, the material is of sufficient purity for the macrocyclization reaction.

A standard procedure for the synthesis of tether T28 is shown in FIG.  ${\bf 10}$ .

Henry reaction of 2-hydroxybenzaldehyde 28-0 provided 28-1 in 79% yield. This was followed by reduction first with sodium borohydride, then with catalytic hydrogenation, to 25 give the amine, which was then protected as its Boc derivative, 28-2. Yields of these first two steps were lower on larger scales. Alkylation of 28-2 with the TBDMS ether of 2-bromoethanol, itself synthesized by standard methods, gave 28-3 in 74% yield. Deprotection of the silyl ether under standard conditions yielded the desired protected tether, Boc-T28. Alternative use of ethylene carbonate for the phenol alkylation to avoid the protection/deprotection steps, gave 73% yield.

A standard procedure for the synthesis of tether T32 is shown in FIG. 11.

A standard procedure for the synthesis of tether T33a and T33b is shown in FIGS. 12A and 12B.

The construction to the (R)-isomer of this tether (T33a)  $_{40}$  was accomplished from 2-iodophenol (33-0) and (S)-methyl lactate (33-A). Mitsunobu reaction of 33-0 and 33

A proceeded with inversion of configuration in excellent yield to give 33-1. Reduction of the ester to the corresponding alcohol (33-2) also occurred in high yield and was followed 45 by Sonagashira reaction with Ddz-propargylamine. The alkyne in the resulting coupling product, 33-3, was reduced with catalytic hydrogenation. Workup with scavenger resin provided the desired product, Ddz-T33a.

The synthesis of the (S)-enantiomer (Ddz-T33b) was carried out in an identical manner in comparable yield starting from (R)-methyl lactate (33-B). See FIG. **12**B.

Standard procedures for the synthesis of various tethers are shown in the figures: tether T34 (FIG. 13), tether T35 (FIG. 14), tether T36 (FIG. 15), tether T37 (FIG. 16), tether T38 (FIG. 17), tether T39 (FIG. 18), tether T40 (FIG. 19), tether T41 (FIG. 20) and tether T42 (FIG. 21).

E. Examples of Synthetic Strategies for the Macrocyclic Compounds of the Invention

FIG. 22 presents a scheme depicting a thioester strategy for macrocyclic compounds of the present invention.

It should be noted that one or more of the amino acids indicated can be replaced by corresponding hydroxy acids and coupled to the next building block utilizing methods known to those in the art.

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cycle. In this manner, acylation, sulfonylation, alkylation (via reductive amination), guanidine and urea formation were performed via standard methods.

Table 1, hereinbelow, shows a representative, but by no means exclusive, summary of the chemical synthesis of several representative compounds of the invention.

TABLE 1

	Synthesis of Representative Compounds of the Present Invention							
		Synthesis of Represen	uative Compound:	s of the Pres				
	$AA_1$	$AA_2$	$AA_3$	Tether	Tether Attachment	Additional Steps		
1	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Nva	Ddz-T8	Example 29	none		
2	Bts-D-Phe	Boc-D-Val	Boc-Nva	Boc-T8	Example 29	none		
3	Bts-D-Phe	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none		
4	Bts-D-	Boc-D-Val	Boc-Nva	Ddz-T9	Example 29	none		
5	Tyr(tBu) Bts-D- Tyr(tBu)	Boc-D-Ala	Boc-Nva	Ddz-T8	Example 29	none		
6	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Met	Ddz-T8	Example 29	none		
7	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Nle	Ddz-T8	Example 29	none		
8	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Phe	Ddz-T8	Example 29	none		
9	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Val	Ddz-T8	Example 29	none		
10	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Leu	Ddz-T9	Example 29	none		
11	Bts-D-2-Nal	Boc-D-Val	Boc-Nva	Boc-T8	Example 29	none		
12	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Abu	Ddz-T8	Example 29	none		
13	Bts-D-Phe	Boc-D-Val	Boc-Leu	Boc-T9	Example 29	none		
14	Bts-D-2-Nal	Boc-D-Val	Boc-Leu	Boc-T9	Example 29	none		
15	Bts-D- Phe(3Cl)	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none		
16	Bts-D- Phe(4Cl)	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none		
17	Bts-D- Trp(Boc)	Boc-D-Val	Boc-Nva	Ddz-T9	Example 29	none		
18 19	Bts-D- Tyr(tBu) Bts-D-	Boc-D-2-Abu Boc-D-Val	Boc-Nva	Ddz-T9 Boc-T9	Example 29	none		
20	Phe(4F) Bts-D-Phe	Boc-D-Val	Boc-Nva Boc-Leu	Вос-Т8	Example 29 Example 29	none		
21	Bts-D-2-Nal	Boc-D-Val	Boc-Leu Boc-Leu	Boc-T8	Example 29	none		
22	Bts-D- Tyr(OMe)	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none		
23	Bts-D-1-Nal	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none		
24	Bts-D-2-Thi	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none		
25	Bts-D- Phe(2Cl)	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none		
26	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Cpa	Ddz-T9	Example 29	none		
27	Bts-D-4-Thz	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none		
28	Bts-D-3-Pal	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none		
29	Bts-D- Tyr(tBu)	Boc-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	none		
30	Bts-D- Tyr(tBu)	Hnva(THP)	Boc-Nva	Ddz-T9	Example 29	none		
34	Bts-D- Tyr(tBu)	Ddz-D-Tyr(tBu)	Boc-Nva	Ddz-T8	Example 29	None		
38	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Ala	Ddz-T8	Example 29	none		
39	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Ala	Ddz-T8	Example 29	none		
40	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Gly	Ddz-T8	Example 29	none		
41	Bts-D- Tyr(tBu)	Boc-DPhe	Boc-Nva	Ddz-T8	Example 29	none		
52	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Phg Ddz-Lys(Boc)	Ddz-T8	Example 29	none		
55	Bts-D- Tyr(tBu)	Ddz-D-Val		Ddz-T8	Example 29	none		
56 57	Bts-D- Tyr(tBu)	Ddz-D-Val	Ddz-Orn(Boc)	Ddz-T8	Example 29	none		
57	Bts-D- Tyr(tBu)	Ddz-D-Val	Ddz-Ser(tBu)	Ddz-T8	Example 29	none		

TABLE 1-continued

		Synthesis of Represe	entative Compounds		Invention	
	$AA_1$	$AA_2$	AA <sub>3</sub>	Tether	Tether Attachment	Additional Steps
	-					-
58	Bts-D- Tyr(tBu)	Ddz-D-Val	Ddz-Tyr(tBu)	Ddz-T8	Example 29	none
59	Bts-D-	DdzD-Val	Ddz-Trp(Boc)	Ddz-T8	Example 29	none
60	Tyr(tBu) Bts-D-	Boc-D-Val	Boc-Tyr(OMe)	Ddz-T8	Example 29	none
65	Tyr(tBu) Bts-D-	Boc-D-Val	Boc-Nva	Ddz-T2	Example 29	none
03	Tyr(tBu)				-	none
71	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Nva	Ddz-T10	Example 29	none
72	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-2-Nal	Ddz-T8	Example 29	none
76	Bts-D-	Boc-D-2-Nal	Boc-Nva	Ddz-T8	Example 29	none
77	Tyr(tBu) Bts-D-	Boc-D-Nle	Boc-Nva	Ddz-T8	Example 29	none
80	Tyr(tBu) Bts-D-	Boc-D-Val	Boc-Ile	Ddz-T8	Example 29	none
85	Tyr(tBu) Bts-D-	Boc-D-Val	Boc-D-Nva	Ddz-T8	Example 29	none
87	Tyr(tBu) Bts-D-Bip	Boc-D-Val	Boc-Nva	Вос-Т9	Example 29	none
	Bts-D-	Boc-D-Val	Boc-Nva	Ddz-T9	Example 29	none
89	Tyr(tBu) Bts-D-Hfe	Boc-D-Val	Boc-Nva	Вос-Т9	Example 29	none
	Bts-D-Dip	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none
	Bts-D-	Boc-D-Nva	Boc-Nva	Ddz-T9	Example 29	none
92	Tyr(tBu) Bts-D-	Boc-D-Tle	Boc-Nva	Ddz-T9	Example 29	none
96	Tyr(tBu) Bts-D-	Boc-β-Ala	Boc-Nva	Ddz-T9	Example 29	none
97	Tyr(tBu) Bts-D- Tyr(tBu)	Boc-D-Chg	Boc-Nva	Ddz-T9	Example 29	none
98	Bts-D-	Boc-D-Val	Boc-Nva	Ddz-T18	Example 29	none
99	Tyr(tBu) Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Nva	Ddz-T15	Example 29	none
109	Bts-D- Tyr(tBu)	Boc-D-Val	Ddz-Dab(Boc)	Ddz-T9	Example 29	none
110	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Nva	Ddz-T11	Example 29	none
111	Bts-D- Tyr(tBu)	Boc-D-Val	Hval(THP)	Ddz-T9	Example 29	none
112	Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Nva	Ddz-T9	Example 29	none
120	Bts-D-	Boc-D-Pro	Boc-Nva	Ddz-T8	Example 29	none
121	Tyr(tBu) Bts-D- Tyr(tBu)	Boc-D-Val	Boc-Nva	Ac-T8-NH2	Example 29	none
122	Boc-D-3-Pal	Boc-D-Val	Boc-Nva	Boc-T9	Example 30	none
123	Boc-D-2-Pal	Boc-D-Val	Boc-Nva	Boc-T9	Example 30	none
	Boc-D-4-Pal	Boc-D-Val	Boc-Nva	Вос-Т9	Example 30	none
	Bts-D-	Boc-D-Cpg	Boc-Nva	Boc-T9	Example 29	none
126	Tyr(tBu) Bts-D- Tyr(tBu)	Boc-D-Val	Boc-NMeLeu	Вос-Т9	Example 29	none
127	Boc-D- His(Mts)	Boc-D-Val	Boc-Nva	Boc-T12	Example 30	none
128	Bts-D- Tyr(OMe)	Boc-D-Val	Boc-Leu	Вос-Т9	Example 29	none
129	Bts-D-1-Nal	Boc-D-Val	Boc-Leu	Boc-T9	Example 29	none
	Bts-D-2-Thi	Boc-D-Val	Boc-Leu	Boc-T9	Example 29	none
131	Bts-D- Phe(3Cl)	Boc-D-Val	Boc-Leu	Boc-T9	Example 29	none
132	Bts-D- Phe(4Cl)	Boc-D-Val	Boc-Leu	Boc-T9	Example 29	none
133	Bts-D- Phe(4F)	Boc-D-Val	Boc-Leu	Вос-Т9	Example 29	none
134	Bts-D- Phe(3Cl)	Boc-D-Val	Boc-Leu	Boc-T2	Example 29	none
135	Bts-D- Tyr(OMe)	Boc-D-Val	Boc-Leu	Boc-T11	Example 29	none

TABLE 1-continued

	Sy	nthesis of Represen	tative Compounds	s of the Present In	vention	
	$AA_1$	$AA_2$	$AA_3$	Tether	Tether Attachment	Additional Steps
136	Bts-D-1Nal	Boc-D-Val	Boc-Leu	Boc-T11	Example 29	none
137	Bts-D-2-Thi	Boc-D-Val	Boc-Leu	Boc-T11	Example 29	none
138	Bts-D-	Boc-D-Val	Boc-Leu	Boc-T11	Example 29	none
139	Phe(3Cl) Bts-D-	Boc-D-Val	Boc-Leu	Boc-T11	Example 29	none
140	Phe(4Cl) Bts-D- Phe(4F)	Boc-D-Val	Boc-Leu	Boc-T11	Example 29	none
141	Bts-D- Tyr(OMe)	Boc-D-Val	Вос-Сра	Вос-Т9	Example 29	none
142	Bts-D-1-Nal	Boc-D-Val	Boc-Cpa	Boc-T9	Example 29	none
143	Bts-D-2-Thi	Boc-D-Val	Boc-Cpa	Boc-T9	Example 29	none
144	Bts-D-	Boc-D-Val	Boc-Cpa	Boc-T9	Example 29	none
145	Phe(3Cl) Bts-D- Phe(4Cl)	Boc-D-Val	Boc-Cpa	Boc-T9	Example 29	none
146	Bts-D- Phe(4F)	Boc-D-Val	Boc-Cpa	Вос-Т9	Example 29	none
147	Bts-D- Tyr(OMe)	Boc-D-Val	Boc-Cpa	Boc-T11	Example 29	none
148	Bts-D-1-Nal	Boc-D-Val	Boc-Cpa	Boc-T11	Example 29	none
149	Bts-D- Phe(3Cl)	Boc-D-Val	Boc-Cpa	Boc-T11	Example 29	none
	Bts-D- Phe(4Cl)	Boc-D-Val	Boc-Cpa	Boc-T11	Example 29	none
	Bts-D- Phe(4F)	Boc-D-Val	Boc-Cpa	Boc-T11	Example 29	none
	Bts-D- Tyr(OMe)	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	none
	Bts-D-1-Nal	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9 Ddz-T9	Example 29	none
	Bts-D-2-Thi Bts-D-	Ddz-D-Val Ddz-D-Val	Ddz-Dap(Boc) Ddz-Dap(Boc)	Ddz-19 Ddz-T9	Example 29 Example 29	none none
	Phe(3Cl) Bts-D-	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	none
	Phe(4Cl) Bts-D-	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	none
158	Phe(4F) Bts-D-	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T11	Example 29	none
159	Phe(3Cl) Bts-D-	Boc-D-Ile	Boc-Nva	Вос-Т9	Example 29	none
160	Tyr(But) Bts-D-	Boc-D-allolle	Boc-Nva	Вос-Т9	Example 29	none
161	Tyr(But) Boc-D- Phe(4CH2NHFmoc)	Boc-D-Val	Boc-Nva	Вос-Т9	Example 30	none
162	Bts-D- Phe(2Me)	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none
163	Bts-D- Phe(3Me)	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none
164	Bts-D- Phe(4Me)	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none
	Bts-D- Phe(3OMe)	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none
	Bts-D- Phe(2OMe)	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none
	Bts-D-3- benzothienyl	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none
	Bts-D-3-Thi Bts-D- HomoPhe(3Cl)	Boc-D-Val Boc-D-Val	Boc-Nva Boc-Nva	Boc-T9 Boc-T9	Example 29 Example 29	none none
170	Bts-D- Phe(3,4diCl)	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none
171	Bts-D- Phe(3,4diF)	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none
172	Bts-D- Phe(3,4diOMe)	Boc-D-Val	Boc-Nva	Вос-Т9	Example 29	none
	Bts-D-1Nal Bts-D-	Hnva(THP) Hnva(THP)	Boc-Nva Boc-Nva	Boc-T9 Boc-T9	Example 29 Example 29	none none
175	Tyr(OMe) Bts-D-	Boc-D-Val	Boc-Nva	Boc-T33b	Example 29	none
176	Tyr(tBu) Bts-D-	Boc-D-Val	Boc-Nva	Boc-T33a	Example 29	none
177	Tyr(tBu) Bts-D-	Boc-D-Val	Boc-Nva	Boc-T28	Example 29	none
	Tyr(tBu)					

TABLE 1-continued

	TIBBE I COMMING							
		Synthesis of Repres	entative Compound	s of the Present Ir	vention			
	$AA_1$	$AA_2$	$AA_3$	Tether	Tether Attachment	Additional Steps		
178	Bts-D- Tyr(OMe)	Ddz-D-Val	Ddz-Ser(tBu)	Ddz-T9	Example 29	none		
179	Bts-D-1-Nal	Ddz-D-Val	Ddz-Ser(tBu)	Ddz-T9	Example 29	none		
180	Bts-D-2-Thi	Ddz-D-Val	Ddz-Ser(tBu)	Ddz-T9	Example 29	none		
181	Bts-D-	Ddz-D-Val	Ddz-Ser(tBu)	Ddz-T9	Example 29	none		
182	Phe(3Cl) Bts-D- Phe(4Cl)	Ddz-D-Val	Ddz-Ser(tBu)	Ddz-T9	Example 29	none		
183	Bts-D- Phe(4F)	Ddz-D-Val	Ddz-Ser(tBu)	Ddz-T9	Example 29	none		
184	Bts-D-1-Nal	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T11	Example 29	none		
185	Bts-D- Phe(4Cl)	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T11	Example 29	none		
186	Ddz-D- Tyr(tBu)	Ddz-D-Val	Ddz-His(Mts)	Ddz-T9	Example 30	none		
187	Bts-D- Phe(3CF3)	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none		
188	Bts-D- Phe(3F)	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none		
189	Bts-D- Phe(4NO2)	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none		
190	Bts-D-3- benzothienyl	Boc-D-Val	Boc-Cpa	Boc-T9	Example 29	none		
191	Bts-D- Phe(3OMe)	Boc-D-Val	Boc-Cpa	Вос-Т9	Example 29	none		
192	Bts-D- Phe(3,4diCl)	Boc-D-Val	Boc-Cpa	Вос-Т9	Example 29	none		
193	Bts-D- Phe(3,4diF)	Boc-D-Val	Boc-Cpa	Вос-Т9	Example 29	none		
194	Bts-D- Tyr(OMe)	Boc-D-Val	Boc-Nva	Boc-T34	Example 29	none		
195	Bts-D- Tyr(OMe)	Boc-D-Val	Boc-Nva	Boc-T38	Example 29	none		
196	Bts-D- Phe(3Cl)	Boc-D-Val	Boc-Cpa	Ddz-T32(Boc)	Example 29	none		
197	Bts-D- Phe(3Cl)	Boc-D-Val	Boc-Cpa	Boc-T34	Example 29	none		
198	Bts-D- Phe(3Cl)	Boc-D-Val	Boc-Cpa	Boc-T38	Example 29	none		
199	Bts-D- Phe(3Cl)	Boc-D-Val	Boc-Cpa	Boc-T41	Example 29	none		
200	Bts-D- Phe(3Cl)	Boc-D-Val	Boc-Cpa	Boc-T8	Example 29	none		
	Bts-D-1-Nal	Boc-D-Val	Boc-Nva	Boc-T8	Example 29	none		
202	Bts-D- Phe(3OMe)	Boc-D-Val	Boc-Nva	Boc-T8	Example 29	none		
203	Bts-D- Phe(4Cl)	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	acetylation		
204	Bts-D- Phe(4Cl)	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	guanidinylation		
205	Bts-D- Phe(3Cl)	Boc-D-Val	Boc-NMeLeu	Boc-T9	Example 29	none		
206	Bts-D- Phe(4Cl)	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	mesylation		
207		Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	TMS- isocyanate followed by dilute acid		
208	Bts-D- Tyr(tBu)	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	guanidinylation		
209	Bts-D- Tyr(tBu)	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	acetylation		
210	Bts-D- Tyr(tBu)	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	reductive amination with acetone		
211	Bts-D- Phe(4Cl)	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	reductive amination with excess formaldehyde		
	Bts-D- Phe(4Cl)	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	reductive amination with acetone		
213	Bts-D- Tyr(3,5dil)	Boc-D-Val	Boc-Nva	Вос-Т9	Example 29	none		

TABLE 1-continued

	Synthesis of Representative Compounds of the Present Invention						
	$AA_1$	$AA_2$	$AA_3$	Tether	Tether Attachment	Additional Steps	
214	Bts-D- Tyr(OMe)	Boc-D-Val	Boc-Hse(Bzl)	Вос-Т9	Example 29	hydrogenolysis for protecting group removal	
215	Bts-D- Tyr(tBu)	Ddz-D-Val	Ddz-Dap(Boc)	Ddz-T9	Example 29	reductive amination with excess formaldehyde	
216	Bts-D- Phe(3Cl)	Boc-D-Val	Boc-Cpa	Boc-T40	Example 29	none	
217	Bts-D- Phe(3Cl)	Boc-D-Val	Boc-Cpa	Boc-T36	Example 29	none	
218	Bts-D- Phe(3Cl)	Boc-D-Val	Boc-Nva	Boc-T39	Example 29	none	
219	Bts-D- Phe(3Cl)	Boc-D-Val	Boc-Nva	Boc-T37	Example 29	none	
220	` /	Boc-D-Val	Boc-Nva	Boc-T39	Example 29	none	
221		Boc-D-Val	Boc-Nva	Boc-T35	Example 29	none	
222		Boc-D-Val	Boc-Nva	Boc-T9	Example 29	none	
223	Bts-D- Tyr(But)	Boc-D-Val	Boc-Nva	Boc-T9	Example 29	acetylation	
224		Boc-D-Val	Boc-Leu	Boc-T9	Example 29	reductive amination with formaldehyde	
225	Bts-D-1-Nal	Boc-D-Val	Boc-Leu	Boc-T9	Example 29	acetylation	
226	Bts-D-1-Nal	Boc-D-Val	Boc-Leu	Boc-T9	Example 29	reductive amination with aldehyde	
227	Bts-D-1-Nal	Boc-D-Val	Boc-Leu	Вос-Т9	Example 29	reductive amination with benzaldehyde	

Notes

Any amino acid or tether designated as the Boc derivative could be substituted with the corresponding Ddz derivative

# D. Analytical Data for Selected Compounds of the Invention

<sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on a Varian Mer- <sup>40</sup> cury 300 MHz spectrometer (Varian, Inc., Palo Alto, Calif.) and are referenced internally with respect to the residual proton signals of the solvent. Information about the conformation of the molecules in solution can be determined utilizing appropriate two-dimensional NMR techniques known to those skilled in the art. HPLC purifications were run on a Waters Xterra® MS C18 column, using the Waters Fraction-Lynx® system (Waters Corporation, Milford, Mass.). Automated medium pressure chromatographic purifications were 50 performed on an Isco CombiFlash® 16× system with disposable silica or C18 cartridges that permitted up to sixteen (16) samples to be run simultaneously (Teledyne Isco, Inc., Lincoln, Nebr.). MS spectra were recorded on a Waters Micromass® Platform II or ZQTM system. HRMS spectra were 55 recorded with a VG Micromass ZAB-ZF spectrometer. Chemical and biological information were stored and analyzed utilizing the ActivityBase® database software (ID Business Solutions Ltd., Guildford, Surrey, UK). General Methods for Analytical HPLC Analyses

HPLC analyses are performed on a Waters Alliance® system 2695 running at 1 mL/min using an Xterra MS C18 column 4.6×50 mm (3.5  $\mu m$ ). A Waters 996 PDA provided UV data for purity assessment (Waters Corporation, Milford, Mass.). An LCPackings (Dionex Corporation, Sunnyvale, 65 Calif.) splitter (50:40:10) allowed the flow to be separated in three parts. The first part (50%) went to a Micromass® Plat-

form II MS equipped with an APCI probe for identity confirmation. The second part (40%) went to an evaporative light scattering detector (ELSD, Polymer Laboratories, now part of Varian, Inc., Palo Alto, Calif., PL-ELS-1000™) for purity assessment and the last portion (10%) to a chemiluminescence nitrogen detector (CLND, Antek® Model 8060, Antek Instruments, Houston, Tex., part of Roper Industries, Inc., Duluth, Ga.) for quantitation and purity assessment. Data was captured and processed utilizing the most recent version of the Waters Millenium® software package (Milford, Mass.).

An example LC method suitable for compounds of the present invention uses MeOH as solvent A,  $\rm H_2O$  as solvent B and 1% TFA/ $\rm H_2O$  as solvent D. Initial mobile-phase composition is 5% A, 85% B and 10% D. Details of the standard gradient method are shown below:

5	Time	Α%	В%	D %	Curve
	0.00	5	85	10	6
	1.00	5	85	10	6
	6.00	50	40	10	6
	9.00	50	40	10	6
0	14.00	90	0	10	6
	17.00	90	0	10	6
	17.50	5	85	10	6
	20.00	5	85	10	6

Compounds 2-6, 8-10, 56, 65 and 144 are as defined in Table (3), hereinbelow.

Compound 2

Yield: 12 mg pure macrocycle was obtained (CLND quan-

<sup>1</sup>H NMR (300 MHz, DMSO-d<sub>6</sub>)  $\delta$  8.83 (m, 1H); 8.53 (m, 1H); 7.63 (m, 1H); 7.4-7.08 (m, 7H); 7.00-6.84 (m, 2H); 6.60 5 (d, 15 Hz, 1H); 6.41 (dt, 15 Hz, 5.4 Hz, 1H); 4.35 (m, 1H); 4.25-4.05 (m, 3H); 3.94 (dt, 1H, 6 Hz, 15 Hz); 3.79 (dd, 1H, 3.6 Hz, 8.4 Hz); 3.60 (m, 1H); 3.52-3.40 (bd, 1H); 3.22-3.06 (m, 4H); 1.88 (m, 2H); 1.54-1.28 (m, 2H); 1.25 (d, 3H, 4.8 Hz); 1.22 (d, 3H, 2.7 Hz); 0.92-0.80 (m, 6H).

HRMS calc. for  $C_{30}H_{40}N_4O_4$ : 520.3049. found 520.3057±0.0016.

HPLC [standard gradient method (refers to that presented in General Methods for Analytical HPLC Analyses)]  $t_R$ =9.55

Compound 4

Yield: 12 mg pure macrocycle was obtained (CLND quantification).

<sup>1</sup>H NMR (300 MHz, DMSO-d<sub>6</sub>) δ 9.35 (b, 1H); 8.98 (b, 1H); 5.52 (d, 1H, 8.4 Hz); 8.38 (b, 1H); 7.25 (b, 1H); 7.13- 20 7.07 (m, 4H); 6.86 (t, 2H, 7.5 Hz); 6.57 (d, 2H, 8.7 Hz); 4.33 (b, 1H); 4.21-4.02 (m, 3H); 3.78 (dd, 1H, 3.3 Hz; 8.1 Hz); 3.65-3.54 (m, 1H); 3.31-3.23 (m, 1H); 3.13-3.02 (m, 4H); 2.78-2.2.28-2.18 (m, 1H); 2.0-1.80 (m, 2H); 1.50-1.30 (m, 3H); 1.25 (d, 3H, 4.5 Hz); 1.22 (d, 3H, 4.5 Hz); 1.01 (d, 3H, 25 6.6 Hz); 0.90 (d, 3H, 6.6 Hz); (t, 3H, 7.5 Hz).

<sup>13</sup>C NMR (75.5 MHz, DMSO-d<sub>6</sub>) δ 172.22; 171.37; 157.77; 157.44; 156.04; 131.76; 130.80; 130.70; 127.88; 121.82; 115.83; 111.71; 62.13; 60.62; 54.21; 52.81; 47.13; 42.47; 33.31; 29.69; 29.30; 28.61; 20.36; 19.44; 18.72; 17.60; 30 13.97.

HRMS calc. for  $C_{30}H_{42}N_4O_5$ : 538.3155. found: 538.3145±0.0016.

HPLC (standard gradient)  $t_R$ =8.12 min.

Compound 5

Yield: 17 mg pure macrocycle was obtained (CLND quantification).

<sup>1</sup>H NMR (300 MHz, DMSO-d<sub>6</sub>) δ 9.02 (b, 1H); 8.47 (d, 1H, 8.4 Hz); 7.7 (b, 1H); 7.58 (d, 1H, 5.4 Hz); 7.28 (dd, 1H, Hz); 6.98-6.91 (m, 3H); 6.66 (d, 8.7 Hz); 6.63 (d, 1H, 15.0 Hz); 6.43 (dt, 1H, 6.0 Hz, 15.0 Hz); 4.28-3.86 (m, 6H); 3.60-3.40 (m, 2H); 3.22-3.12 (m, 1H0; 3.05 (d, 2H, 5.4 Hz); 1.92-1.80 (m, 1H); 1.56-1.40 (m, 1H); 1.36-1.20 (m, 2H); 1.25 (d, 3H, 6.6 Hz); 0.84 (t, 3H, 7.2 Hz).

<sup>13</sup>C NMR (75.5 MHz, DMSO-d<sub>6</sub>) δ 172.54; 171.86; 158.97; 158.56; 127.39; 155.84; 131.62; 129.73; 129.20; 129.02; 128.43; 126.30; 124.51; 122.01; 115.85; 112.88; 61.23; 52.90; 51.23; 47.08; 42.66; 36.13; 33.30; 21.14; 19.57; 17.07; 14.14; 11.49.

HRMS calc. for  $C_{28}H_{36}N_4O_5$ : 508.2685. found: 508.2681±0.0015.

HPLC (standard gradient)  $t_R$ =7.67 min.

Compound 6

Yield: 16 mg pure macrocycle was obtained (CLND quan- 55 tification).

<sup>1</sup>H NMR (300 MHz, DMSO-d<sub>6</sub>) δ 9.37 (b, 1H); 8.87 (b, 1H); 8.61 (d, 1H, 8.7 Hz); 7.62 (b, 1H); 7.27 (d, 1H, 7.8 Hz); 7.21 (t, 1H, 8.4 Hz); 7.14 (d, 2H, 8.4 Hz); 6.98-6.87 (m, 3H); 6.64 (d, 2H, 8.1 Hz); 6.70 (d, 1H, 15.6 Hz); 6.39 (dt, 1H, 6.3 60 Hz, 15.6 Hz); 4.44-4.36 (m, 1H); 4.34-4.08 (m, 2 Hz); 4.45-3.92 (dt, 1H, 6.9 Hz, 15.6 Hz); 3.74 (dd, 1H, 3.6 Hz, 8.4 Hz); 3.54-3.26 (m, 3H); 3.22-3.02 (m, 3H); 2.60-2.36 (m, 4H); 2.24-2.14 (m, 1H); 2.02 (s, 3H); 1.96-1.89 (m, 1H); 1.80-1.66 (m, 1H); 1.01 (d, 3H, 6.3 Hz); 0.90 (d, 3H, 6.6 Hz).

<sup>13</sup>C NMR (75.5 MHz, DMSO-d<sub>6</sub>) δ 171.51; 171.26; 158.90; 158.49; 157.38; 155.86; 131.63; 129.82; 129.21;

128.86; 128.63; 126.21; 121.98; 115.83; 112.83; 62.11; 61.06; 51.97; 47.10; 42.78; 30.91; 30.67; 29.34; 20.37; 19.39; 15.06.

HRMS calc. for  $C_{30}H_{40}N_4O_5S$ : 568.2719. found: 568.2711±0.0017.

HPLC R, (general method) 7.92 min.

Compound 8

Yield: 27 mg pure macrocycle was obtained (CLND quantification).

<sup>1</sup>H NMR (300 MHz, DMSO-d<sub>6</sub>) δ 9.05 (b, 1H); 8.43 (b, 1H); 8.34 (d, 1H, 9.3 Hz); 7.40 (b, 1H); 6.97 (d, 1H, 7.5 Hz); 6.92-6.74 (m, 9H); 6.67-6.54 (m, 2H); 6.33-6.25 (m, 3H); 6.10 (dt, 1H, 5.7 Hz, 16.2 Hz); 4.22 (dt, 1H, 0.9 Hz, 12 Hz); 3.94-6.66 (m, 4H); 3.30 (dd, 1H, 3.6 Hz, 7.8 Hz); 3.24 (m, 1H); 3.18 (m, 1H); 2.85-2.68 (m, 3H); 2.44-2.23 (m, 2H); 1.32 (o, 1H, 7.5 Hz); 0.97-0.89 (m, 1H); 0.42 (d, 3H, 6.6 Hz); 0.01 (d, 3H, 6.6 Hz).  $^{13}$ C NMR (75.5 MHz, DMSO-d<sub>6</sub>)  $\delta$ 171.20; 157.35; 155.88; 139.12; 131.61; 130.87; 129.74; 129.21; 128.77; 128.88; 126.85; 126.19; 121.97; 115.82; 112.84; 62.04; 61.10; 55.07; 50.01; 47.09; 42.85; 37.42; 29.11.

HRMS calc. For  $C_{34}H_{42}N_4O_5$ : 586.3155. found: 586.3145±0.0017.

HPLC R<sub>t</sub> (general method) 9.34 min. Compound 9

Yield: 17 mg pure macrocycle was obtained (CLND quantification).

<sup>1</sup>H NMR (300 MHz, DMSO-d<sub>6</sub>) δ 9.39 (b, 1H); 8.83 (b, 1H); 8.29 (d, 1H, 9.3 Hz); 7.62 (b, 1H); 7.28 (d, 1H, 6.6 Hz); 7.20 (t, 1H, 6.9 Hz); 7.12 (d, 2H, 7.8 Hz); 6.98-6.91 (m, 2H); 6.63 (d, 2H, 8.4 Hz); 6.58 (d, 1H, 16.2 Hz); 6.40 (dt, 1H, 5.7 Hz, 16.2 Hz); 4.29-4.13 (m, 3H); 4.03-3.92 (m, 2H); 3.52 (m, 35 1H); 3.15-3.05 (m, 3H); 2.45-2.37 (m, 1H);

1.96-1.88 (m, 1H); 1.25 (dd, 2H, 4.5 Hz; 6 Hz); 1.01 (d, 3H, 6.3 Hz); 0.91 (d, 3H, 6.6 Hz); 0.86 (d, 3H, 7.2 Hz); 0.81 (d, 3H, 6.6 Hz).

<sup>13</sup>C NMR (75.5 MHz, DMSO-d<sub>6</sub>) δ 171.85; 171.17; 7.8~Hz, 0.8~Hz); 7.20~(t, 1H, 9.0~Hz, 0.8~Hz); 7.14~(d, 2H, 8.4~~40~~157.37;~155.87;~131.59;~129.88;~129.18;~128.97;~128.78;~129.18;128.51: 126.16: 121.97: 115.83: 112.85: 61.55: 61.18: 58.15: 54.22; 47.08; 42.89; 36.32; 29.35; 29.00; 20.34; 19.56; 18.73; 17.44.

> HRMS calc. for  $C_{30}H_{40}N_4O_5$  536.2998. found: 45 536.2990±0.0017.

HPLC (standard gradient)  $t_R$ =8.15 min.

Compound 10

Yield: 24 mg pure macrocycle was obtained (CLND quantification).

<sup>1</sup>H NMR (300 MHz, DMSO-d<sub>6</sub>) δ 9.33 (b, 1H); 8.82 (b, 1H); 8.56 (d, 1H, 8.3 Hz); 7.60 (b, 1H); 7.27 (d, 2H, 7.8 Hz); 7.20 (t, 1H, 7.8 Hz); 7.13 (d, 2H, 8.4 Hz); 6.95 (t, 2H, 7.8 Hz); 6.64 (d, 2H, 8.4 Hz); 6.57 (d, 1H, 15.4 Hz); 6.38 (dt, 1H, 15.4 Hz, 5.8 Hz); 4.26-4.10 (m, 3H); 3.96 (dt, 1H, 5.4 Hz, 8.4 Hz); 3.77 (dd, 1H, 3.7 Hz, 7.8 Hz); 3.51-3.24 (m, 3H); 3.18-3.02 (m, 3H); 1.90 (h, 1H, 6.4 Hz); 1.73-1.54 (m, 2H); 1.45 (dt, 1H, 6.7 Hz, 0.9 Hz); 0.99 (d, 3H, 6.6 Hz); 0.89 (d, 3H, 6.3 Hz); 0.87 (d, 3H, 6.0 Hz); 0.80 (d, 3H, 6.3 Hz).

<sup>13</sup>C NMR (75.5 MHz, DMSO-d<sub>6</sub>) δ 172.23; 171.17; 157.37; 155.88; 131.62; 129.82; 129.19; 128.95; 128.59; 126.24; 121.99; 115.84; 112.88; 64.23; 61.98; 61.14; 51.43; 61.14; 51.43; 47.07; 42.81; 29.38; 24.85; 24.11; 21.00; 20.32; 19.30.

HRMS calc. for  $C_{31}H_{42}N_4O_5$  550.3155. found: 550.3150±0.0016.

HPLC (standard gradient)  $t_R$ =8.91 min.

Compound 56

Yield: 16 mg pure macrocycle was obtained (CLND quantification).

 $^{1}$ H NMR (300 MHz, DMSO-d<sub>6</sub>) δ 9.39 (b, 1H); 8.90 (b, 1H); 8.67 (d, 1H, 8.4 Hz); 7.74 (b, 4H); 7.29-7.08 (m, 4H); 5 6.99-6.87 (m, 2H); 6.64 (d, 2H, 8.1 Hz); 6.61 (d, 1H, 16.5 Hz); 6.40 (dt, 1H, 5.7 Hz, 16.5 Hz); 4.40-4.06 (m, 4H); 4.02-3.95 (m, 1H); 3.79 (dd, 1H, 3.6 Hz, 7.8 Hz); 3.55-3.30 (m, 2H); 3.16-3.05 (m, 3H); 2.82-2.69 (m, 2H); 2.02-1.85 (m, 2H); 1.64-1.43 (m, 3H); 1.29-1.23 (m, 1H); 1.01 (d, 3H, 6.3 Hz);  $^{10}$ 0.91 (d, 3H, 6.3 Hz); 0.86-0.84 (m, 2H).

HPLC (standard gradient)  $t_R$ =5.71 min.

Compound 65

Yield: 17 mg pure macrocycle was obtained (CLND quantification).

 $^{1}\mathrm{H}$  NMR (300 MHz, DMSO-d<sub>6</sub>)  $\delta$  9.60 (b, 1H); 9.39 (b, 1H); 8.88 (b, 1H); 8.70 (d, 1H, 7.5 Hz); 8.57 (d, 1H, 4.2 Hz); 7.27 (t, 6 Hz); 6.96 (d, 2H, 8.4 Hz); 6.66 (d, 2H, 8.4 Hz); 5.78-5.68 (m, 1H); 5.42-5.33 (m, 1H); 3.96-3.89 (m, 1H); 3.80-3.57 (m, 5H); 3.41-3.34 (m, 1H); 3.10-2.90 (m, 1H); 2.78-2.66 (m, 1H); 2.21-2.10 (m, 1H); 2.06-1.93 (m, 1H); 1.70-1.60 (m, 1H); 1.52-1.41 (m, 1H); 1.39-1.26 (m, 1H); 1.25 (d, 3H, 4.8 Hz); 1.23 (d, 3H, 4.5 Hz); 0.83 (dd, 3H, 3 Hz, 4.5 Hz).

<sup>13</sup>C NMR (75.5 MHz, DMSO-d<sub>6</sub>) δ 172.68; 172.63; <sup>25</sup> 159.15; 158.73; 157.38; 157.25; 130.89; 124.99; 116.03; 62.51; 62.12; 54.29; 49.27; 42.47; 32.77; 30.43; 28.85; 20.46; 19.59; 18.72; 17.39; 13.90; 13.09.

HRMS calc. for  $C_{24}H_{36}N_4O_4$ : 444.2736. found: 444.2726 $\pm$ 0.0013.

HPLC (standard gradient)  $t_R$ =6.80 min.

Compound 144

<sup>1</sup>H NMR (300 MHz, CD<sub>3</sub>OD) δ 7.4 (m, 1H); 7.27 (dt, 1H, 1.5 Hz, 6.6 Hz); 7.22-7.14 (m, 2H); 7.08-6.98 (m, 2H); 6.78 9t, 2H, 6.6 Hz); 4.45-4.39 (m, 2H); 4.15 (d, 2H, 8.1 Hz); 7.74 (d, 1H, 9.3 Hz); 3.54 (d, 1H, 10.8 Hz); 3.35-3.22 (m, 2H); 3.20 (q, 1H, 1.5 Hz); 2.82-2.71 (m, 1H); 2.61-2.55 (m, 1H); 2.21-2.11 (m, 1h); 2.02-1.94 (m, 1H); 1.74-1.40 (m, 5H); 1.04 (d, 3H, 6.6 Hz); 0.93 (d, 3H, 6.6 Hz); 0.74-0.64 9m, 1H); 0.45-0.28 (m, 2H); 0.15-0.08 (m, 1H); 0.06-0.02 (m, 1H).

<sup>13</sup>C NMR (75.5 MHz, CD<sub>3</sub>OD) δ 173.29; 172.14; 167.51; 155.47; 134.86; 134.81; 130.38; 130.31; 128.81; 128.25; 127.44; 121.63; 110.39; 107.71; 105.02; 67.10; 66.66; 62.81; 62.06; 60.10; 53.99; 41.44; 36.07; 31.91; 30.01; 29.18; 28.94; 27.79; 23.68; 23.15; 19.08; 18.25; 8.17; 4.98; 3.16.

HRMS: calc. for  $C_{31}H_{41}N_4O_4C1568.2816$ . found 568.2802±0.0017.

F. Mass Spectral Data for Selected Compounds of the Invention

TABLE 2

Analysis of selected compounds of the invention							
	Molecular Formula	Molecular Weight (calculated)	Monoisotopic Mass	M + H Found			
1	C30H40N4O5	536.7	536	537			
2	C30H40N4O4	520.7	520	521			
3	C30H42N4O4	522.7	522	523	60		
4	C30H42N4O5	538.7	538	539			
5	C28H36N4O5	508.6	508	509			
6	C30H40N4O5S	568.7	568	569			
7	C31H42N4O5	550.7	550	551			
8	C34H42N4O5	586.7	586	587			
9	C30H40N4O5	536.7	536	537	65		
10	C31H42N4O5	550.7	550	551			

TABLE 2-continued

	Analysis of sel	ected compounds of t	he invention	
	Molecular Formula	Molecular Weight (calculated)	Monoisotopic Mass	M + H Found
11	C34H44N4O4	572.7	572	573
12	C29H38N4O5	522.6	522	523
13 14	C31H44N4O4 C35H46N4O4	536.7 586.8	536 586	537 587
15	C30H41N4O4Cl	557.1	556	557
16	C30H41N4O4Cl	557.1	556	557
17	C32H43N5O4	561.7	561	562
18 19	C29H40N4O5 C30H41N4O4F	524.7 540.7	524 540	525 541
20	C31H42N4O4	534.7	534	535
21	C35H44N4O4	584.7	584	585
22	C31H44N4O5	552.7	552	553
23 24	C34H44N4O4 C28H40N4O4S	572.7 528.7	572 528	573 529
25	C30H41N4O4Cl	557.1	556	557
26	C31H42N4O5	550.7	550	551
27	C27H39N5O4S	529.7	529	530
28 29	C29H41N5O4	523.7 525.6	523 525	524 526
30	C28H39N5O5 C30H41N3O6	539.7	539	540
34	C34H40N4O6	600.7	600	601
38	C28H36N4O5	508.6	508	509
39	C28H36N4O5	508.6	508	509
40 41	C27H34N4O5 C34H40N4O5	494.6 584.7	494 584	495 585
52	C33H38N4O5	570.7	570	571
55	C31H43N5O5	565.7	565	566
56	C30H41N5O5	551.7	551	552
57 58	C28H36N4O6 C34H40N4O6	524.6 600.7	524 600	525 601
59	C36H41N5O5	623.7	623	624
60	C35H42N4O6	614.7	614	615
65	C24H36N4O4	444.6	444	445
71 72	C29H40N4O6 C38H42N4O5	540.7 634.8	540 634	541 635
76	C38H42N4O5	634.8	634	635
77	C31H42N4O5	550.7	550	551
80	C31H42N4O5	550.7	550	551
85 87	C30H40N4O5 C36H46N4O4	536.7 598.8	536 598	537 599
88	C34H50N4O5	594.8	594	595
89	C31H44N4O4	536.7	536	537
90	C36H46N4O4	598.8	598	599
91 92	C30H42N4O5 C31H44N4O5	538.7 552.7	538 552	539 553
96	C28H38N4O5	510.6	510	511
97	C33H46N4O5	578.7	578	579
98	C24H39N5O4	461.6	461	462
99 109	C24H39N5O4 C29H41N5O5	461.6 539.7	461 539	462 540
110	C29H41N5O5	539.7	539	540
111	C30H41N3O6	539.7	539	540
112	C31H44N4O5	552.7	552	553
120 121	C30H38N4O5 C32H45N5O6	534.6 595.7	534 595	535 596
121	C32H43N3O0 C31H43N4O4Cl	571.2	570	571
123	C29H41N5O4	523.7	523	524
124	C29H41N5O4	523.7	523	524
125 126	C30H40N4O5 C32H46N4O5	536.7 566.7	536 566	537 567
127	C30H38N6O3S	562.7	562	563
128	C32H46N4O5	566.7	566	567
129	C35H46N4O4	586.8	586	587
130 131	C29H42N4O4S C31H43N4O4Cl	542.7 571.2	542 570	543 571
131	C31H43N4O4Cl	571.2	570	571
133	C31H43N4O4F	554.7	554	555
134	C25H37N4O3Cl	477.0	476	477
135 136	C31H45N5O5 C34H45N5O4	567.7 587.8	567 587	568 588
137	C28H41N5O4S	543.7	543	544
138	C30H42N5O4Cl	572.1	571	572
139	C30H42N5O4Cl	572.1	571	572
140 141	C30H42N5O4F C32H44N4O5	555.7 564.7	555 564	556 565
141	C3211 <del>11</del> 114O3	JU4.7	20 <del>4</del>	505

TABLE 2-continued

 $K_i = level B$ 

	TA	BLE 2-continue	d		TABLE 2-continued					
	Analysis of se	lected compounds of	the invention				Analysis of sel	ected compounds of	the invention	
	Molecular Formula	Molecular Weight (calculated)	Monoisotopic Mass	M + H Found	5	M	olecular Formula	Molecular Weight (calculated)	Monoisotopic Mass	M + H Found
142	C35H44N4O4	584.7	584	585		217 C3	31H40N4O4FCl	587.1	586	587
	C29H40N4O4S	540.7	540	541			31H43N4O4Cl	571.2	570	571
	C31H41N4O4Cl	569.1	568	569			30H40N4O4Cl2	591.6	590	591
	C31H41N4O4Cl	569.1	568	569			31H43N4O4F	554.7	554	555
	C31H41N4O4F C31H43N5O5	552.7 565.7	552 565	553 566	10		30H40N4O4FC1 34H50N4O5	575.1	574 594	575 595
	C34H43N5O4	565.7 585.7	585 585	566 586			32H44N4O6	594.8 580.7	580	595 581
	C30H40N5O4Cl	570.1	569	570			36H48N4O4	600.8	600	601
	C30H40N5O4Cl	570.1	569	570			37H48N4O5	628.8	628	629
	C30H40N5O4F	553.7	553	554			39H49N5O4S	683.9	683	684
152	C29H41N5O5	539.7	539	540	15	227 C4	42H52N4O4	676.9	676	677
	C32H41N5O4	559.7	559	560	13					
	C26H37N5O4S	515.7	515	516		Notes				
	C28H38N5O4Cl	544.1	543	544		Molecula     structure vi	ar formulas and molecu a ActivityBase ® softw	ılar weights (MW) are ca vare (IDBS, Guildford, Su Veight Calculator v. 6.32	lculated automatica	lly from the Wonly, from
	C28H38N5O4Cl	544.1	543	544		the freewar	e program Molecular V	Veight Calculator v. 6.32		
	C28H38N5O4F C27H37N6O4Cl	527.6 545.1	527 544	528 545		2. M + H of	btained from LC-MS a	nalysis using the Genera	l Method as describ	ed
	C31H44N4O5	552.7	552	553	20	3. All analy	ses conducted on mate	rial after preparative HP	LC purification	
	C31H44N4O5	552.7	552	553		Biologi.	cal Methods an	d Reculte		
	C31H45N5O4	551.7	551	552		_			1	. 10
162	C31H44N4O4	536.7	536	537				ne present inventi		
163	C31H44N4O4	536.7	536	537		their ab	ility to interact	at the human mot	ilin receptor ı	ıtilizing
	C31H44N4O4	536.7	536	537	2.5	a comp	etitive radiolig	gand binding as	say as descr	ribed in
	C31H44N4O5	552.7	552	553	25			aracterization of		
	C31H44N4O5 C32H42N4O4S	552.7 578.8	552 578	553 579				functional assay		
	C28H40N4O4S	528.7	528	529				ne of these metho		
	C31H43N4O4Cl	571.2	570	571		-				
	C30H40N4O4Cl2	591.6	590	591				gh throughput n		
171	C30H40N4O4F2	558.7	558	559	30	simulta	neous evaluatio	on of many comp	ounds. Othe	r assays
172	C32H46N4O6	582.7	582	583		have als	so been describ	ed that are suita	ble for HTS,	such as
	C34H43N3O5	573.7	573	574				ble expression o		
	C31H43N3O6	553.7	553	554			nan motilin rece			J
	C31H44N4O5	552.7	552 553	553				nination of repre	sentative com	mounde
	C31H44N4O5 C29H40N4O5	552.7 524.7	552 524	553 525						
	C29H40N4O6	540.7	540	541	35			n using Method		
	C32H40N4O5	560.7	560	561				activity is listed		
	C26H36N4O5S	516.7	516	517				001-0.10 μM; B=		
181	C28H37N4O5Cl	545.1	544	545		10.0 μN	M. In addition,	the assay resul	ts of two ad	ditional
	C28H37N4O5Cl	545.1	544	545		compou	ınds using this	Method are sho	wn below. As	can be
	C28H37N4O5F	528.6	528	529	40	observe	d, this demons	trates the activit	y of a repres	entative
	C31H40N6O4	560.7	560	561	-10			Formula IV of		
	C27H37N6O4Cl C31H40N6O5	545.1 576.7	544 576	545 577				ation of D-proli		
	C31H41N4O4F3	590.7	590	591				k. Significantly,		
	C30H41N4O4F	540.7	540	541						
	C30H41N5O6	567.7	567	568		activity	obtained with	compound 121	, which is th	e imear
190	C33H42N4O4S	590.8	590	591	45	analogu	ie of compound	$11 (K_i = level B),$	illustrates the	critical
	C32H44N4O5	564.7	564	565				elic structure to	attaining the	desired
	C31H40N4O4Cl2	603.6	602	603		interact	ion.			
	C31H40N4O4F2 C32H48N6O6	570.7	570	571						
	C32H46N4O5	612.8 566.7	612 566	613 567						
	C32H43N6O4Cl	611.2	610	611	50				Comp	pound 120
	C32H45N6O5Cl	629.2	628	629	30					
198	C32H43N4O4Cl	583.2	582	583				/ 🖺	TT	
199	C27H39N4O6Cl	551.1	550	551				くた	H N	
	C31H39N4O4Cl	567.1	566	567				\ <sub>N</sub> / \		$\checkmark$
	C34H42N4O4	570.7	570	571						
	C31H42N4O5	550.7	550	551 596	55			-m.,	0	
	C30H40N5O5Cl C29H40N7O4Cl	586.1 586.1	585 585	586 586			J 1	·····	O NI	a
	C32H45N4O4Cl	585.2	584	585		НО		IDI		
	C29H40N5O6SCI	622.2	621	622			_	+N		
	C29H39N6O5Cl	587.1	586	587				]	ĺ	
	C29H41N7O5	567.7	567	568	60			Ĺ		
	C30H41N5O6	567.7	567	568	60			<b>\</b> 0	الر	
	C31H45N5O5	567.7	567	568				\		
	C30H42N5O4Cl C31H44N5O4Cl	572.1 586.2	571 585	572 586				) <del>-</del>	<del>-</del> <	
	C30H40N4O5l2	790.5	790	791				(	)	
	C30H42N4O6	554.7	554	555				//	//	
								v		

553.7

583.2

215 C30H43N5O5

216 C32H43N4O4Cl

Competitive binding curves for two representative compounds of the invention (Compounds 8 and 11) are presented in FIG. 23 and FIG. 24, respectively.

 $K_i > 10 \mu M$ 

For determination of functional significance of the binding, the compounds are preferably tested in the Aequorin assay as described in Method B2, although the procedure of Method B3 is also applicable. As can be seen from the data presented in Table 4, the representative compounds examined act as 30 antagonists at the motilin receptor and are devoid of agonist activity at the concentrations studied. The functional activity is listed as ranges with the following levels: A=0.001-0.10  $\mu$ M; B=0.10-1.0  $\mu$ M. The higher sensitivity of the assay of Method B2, almost 100 times that of Method C, makes it the 35 preferred one for this assessment. This is evident in the  $EC_{50}$ values obtained in each for the positive agonist standard, motilin. Additionally, Method B2 measures the actual signaling event, which makes it more relevant to the effect that is desired, whereas the assay of Method B3 simply measures 40 GTP turnover.

TABLE 4

Demonstration of A	ntagonist Activity at	the Motilin Receptor
	Aequorin (	Method B2) <sup>1</sup>
Compound	Binding $(K_i)$	$IC_{50}$
142	A	В
149	A	В
167	A	A
168	$\mathbf{A}$	A
212	A	A
Motilin (human, porcine) <sup>2</sup>	0.6	not applicable

 $^1Activity$  is listed as ranges with the following levels: A = 0.001-0.10  $\mu M; B$  = 0.10-1.0  $\mu M$   $^2Human$  and porcine motilin are the same peptide.

In addition, a common and scientifically-accepted ex vivo assay for the measurement of agonist or antagonist activity at the motilin receptor is the contraction of rabbit duodenum or other gastrointestinal smooth muscle tissue. A2-A4 Agonists are defined as compounds that induce >50% contraction relative to the motilin peptide, whereas antagonists are defined as compounds that cause >50% inhibition of the response to motilin. Compounds of the present invention have shown 65 significant antagonist activity in this assay. For example, compound 144 exhibited a pA<sub>2</sub>=6.95, while compound 165

had a  $pA_2$ =7.17, as calculated from the Schild plots of the response obtained at various concentrations as described in Method B4.

Gastric motility is generally measured in the clinical setting as the time required for gastric emptying and subsequent transit time through the GI tract. Gastric emptying scans are well known to those skilled in the art an, briefly, comprise use of an oral contrast agent, such as barium, or a radiolabeled meal. Solid and liquids can be measured independently. A test food or liquid is radiolabeled with an isotope (99mTc) and after ingestion or administration, transit time through the GI tract and gastric emptying are measured by visualization using gamma cameras. These studies are performed before and after the administration of the therapeutic agent to quantify the efficacy of the compound.

#### Example Method B1

Competitive Radioligand Binding Assay (Motilin Receptor)

# Materials:

20

Membranes were prepared from CHO cells stably transfected with the human motilin receptor and utilized at a quantity of 1.5 μg/assay point. [PerkinEimer<sup>TM</sup> SignalScreen® Product #6110544, PerkinElmer, Inc., Wellesley, Mass.]

[125]-Motilin (PerkinElmer, #NEX-378); final concentration: 0.04-0.06 nM Motilin (Bachem<sup>TM</sup>, #H-4385, Bachem Bioscience Inc., King of Prussia, Pa.); final concentration: 1 μM

Multiscreen® Harvest plates-GF/B (Millipore™, #MAHFB1H60, Billerica, Mass.)

Deep-well polypropylene titer plate (Beckman Coulter<sup>TM</sup>, #267006, Fullerton, Calif.)

TopSeal-A<sup>TM</sup> (PerkinElmer, #6005185, Wellesley, Mass.) Bottom seal (Millipore<sup>TM</sup>, #MATAH0P00, Billerica, Mass.)

MicroScint-0<sup>TM</sup> (PerkinElmer, #6013611, Wellesley, Mass.)

Binding Buffer: 50 mM Tris-HCl (pH 7.4), 10 mM MgCl<sub>2</sub>, 1 mM EDTA, 0.1% BSA

# Assay Volumes:

150 μL of membranes diluted in binding buffer

10 µL of compound diluted in binding buffer

10  $\mu L$  of radioligand ([ $^{125}I$ ]-Motilin) diluted in binding buffer

Final Test Concentrations (N=11) for Compounds: 10, 5, 2, 1, 0.5, 0.2, 0.1, 0.05, 0.02, 0.01, 0.005  $\mu M$ .

### 50 Compound Handling:

Compounds were provided frozen on dry ice at a stock concentration of 10 mM diluted in 100% DMSO and stored at -20° C. until the day of testing. On the test day, compounds were allowed to thaw at room temperature and than diluted in assay buffer according to the desired test concentrations. Under these conditions, the maximum final DMSO concentration in the assay was 0.5%.

#### Assay Protocol:

In deep-well plates, diluted cell membranes (1.5  $\mu$ g/mL) are combined with 10  $\mu$ L of either binding buffer (total binding, N=5), 1  $\mu$ M motilin (non-specific binding, N=3) or the appropriate concentration of test compound. The reaction is initiated by addition of 10  $\mu$ l of [ $^{125}$ I]-motilin (final conc. 0.04-0.06 nM) to each well. Plates are sealed with TopSeal-A, vortexed gently and incubated at room temperature for 2 hours. The reaction is arrested by filtering samples through pre-soaked (0.3% polyethyleneimine, 2 h) Multiscreen Har-

vest plates using a Tomtec® Harvester (Tomtec, Hamden, Conn.)), washed 9 times with  $500\,\mu\text{L}$  of cold  $50\,\text{mM}$  Tris-HCl (pH 7.4), and than plates are air-dried in a fumehood for 30 minutes. A bottom seal is applied to the plates prior to the addition of  $25\,\mu\text{L}$  of MicroScint- $0^{\text{TM}}$  to each well. Plates are then sealed with TopSeal-A® and counted for 30 sec per well on a TopCount® Microplate Scintillation and Luminescence Counter (PerkinElmer, Wellesley, Mass.) where results are expressed as counts per minute (cpm).

Data are analyzed by GraphPadTM Prism (GraphPad Software, San Diego, Calif.) using a variable slope non-linear regression analysis.  $K_i$  values were calculated using a  $K_d$  value of 0.16 nM for [ $^{125}$ I]-motilin (previously determined during membrane characterization).

test concentration with maximal displacement –  $D_{max} = 1 - \frac{\text{non-specific binding}}{\text{total binding- non-specific binding}} \times 100$ 

where total and non-specific binding represent the cpm obtained in the absence or presence of 1  $\mu$ M motilin, respectively.

#### Example Method B2

Aequorin Functional Assay (Motilin Receptor)

## Materials:

Membranes were prepared using AequoScreen<sup>TM</sup> (EUROSCREEN, Belgium) cell lines expressing the human motilin receptor (cell line ES-380-A; receptor accession #AF034632). This cell line is constructed by transfection of the human motilin receptor into CHO-K1 cells co-expressing  $G_{\alpha 16}$  and the mitochondrially targeted Aequorin (Ref #ES-WT-A5).

Motilin (Bachem<sup>TM</sup>, #H-4385, Bachem Bioscience Inc., King of Prussia, Pa.)

Assay buffer: DMEM-F12 (Dulbeccoe's Modified Eagles Medium) with 15 mM HEPES and 0.1% BSA (pH 7.0) Coelenterazine (Molecular Probes<sup>TM</sup>, Leiden, The Netherlands)

Final Test Concentrations (N=5) for Compounds:

10, 3.16, 1, 0.316, 0.1 μΜ.

Compound Handling:

Compounds were provided as dry films at a quantity of approximately 1.2  $\mu mol$  in pre-formatted 96-well plates. Compounds were dissolved in 100% DMSO at a concentration of 10 mM and stored at  $-20^{\circ}$  C. until further use. Daughter plates were prepared at a concentration of 500  $\mu M$  in 30% DMSO with 0.1% BSA and stored at  $-20^{\circ}$  C. until testing. On 55 the test day, compounds were allowed to thaw at room temperature and than diluted in assay buffer according to the desired test concentrations. Under these conditions, the maximum final DMSO concentration in the assay was 0.6%. Cell Preparation: 60

Cells are collected from culture plates with Ca²+ and Mg²+ free phosphate buffered saline (PBS) supplemented with 5 mM EDTA, pelleted for 2 minutes at  $1000\times g$ , resuspended in assay buffer (see above) at a density of  $5\times10^6$  cells/mL and incubated overnight in the presence of 5  $\mu$ M coelenterazine. 65 After loading, cells were diluted with assay buffer to a concentration of  $5\times10^5$  cells/mL.

54

Assay Protocol:

For agonist testing, 50 µl of the cell suspension was mixed with 50 µl of the appropriate concentration of test compound or motilin (reference agonist) in 96-well plates (duplicate samples). The emission of light resulting from receptor activation was recorded using the Functional Drug Screening System 6000 'FDSS 6000' (Hamamatsu Photonics K.K., Japan).

For antagonist testing, an approximate EC80 concentration of motilin (i.e. 0.5 nM; 100 µL) was injected onto the cell suspension containing the test compounds (duplicate samples) 15-30 minutes after the end of agonist testing and the consequent emission of light resulting from receptor activation was measured as described in the paragraph above.

Results are expressed as Relative Light Units (RLU). Concentration response curves were analyzed using GraphPad<sup>TM</sup> Prism® (GraphPad Software, San Diego, Calif.) by non-linear regression analysis (sigmoidal dose-response) based on the equation  $E=E_{max}/(1+EC50/C)$ n where E is the measured RLU value at a given agonist concentration (C),  $E_{max}$  is the maximal response,  $EC_{50}$  is the concentration producing 50% stimulation and n is the slope index. For agonist testing, results for each concentration of test compound were expressed as percent activation relative to the signal induced by motilin at a concentration equal to the  $EC_{80}$  (i.e. 0.5 nM). For antagonist testing, results for each concentration of test compound were expressed as percent inhibition relative to the signal induced by motilin at a concentration equal to the  $EC_{80}$  (i.e. 0.5 nM).

#### Example Method B3

FlashPlate® Motilin [35S]-GTPyS Functional Assay

Materials

Membranes were prepared from CHO cells stably transfected with the human motilin receptor and utilized at a quantity of 1.5 μg/assay point.

[PerkinElmer Signal/Screen® Product #6110544, PerkinElmer, Inc. Wellesley, Mass.]

GTPγS Guanosine 5'-[γ-thio]triphosphate tetralithium salt (Sigma, #G-8634, Sigma-Aldrich, St. Louis, Mo.)
[35S]-GTPγS (PerkinElmer, #NEX-030H)

Motilin (Bachem<sup>TM</sup>, #H-4385, Bachem Bioscience Inc., King of Prussia, Pa.)

96-well FlashPlate® white polystyrene microplates (PerkinElmer, #SMP200, Wellesley, Mass.)

Deep-well polypropylene titer plate (Beckman Coulter™, #267006, Fullerton, Calif.)

TopSeal-A<sup>TM</sup> (PerkinElmer, #6005185, Wellesley, Mass.) Assay Buffer: 50 mM Tris (pH 7.4), 100 mM NaCl, 10 mM MgCl<sub>2</sub>, 1 mM EDTA, 1  $\mu$ M GDP, 0.1% BSA Assay Volumes:

25 μL of compound diluted in assay buffer

25 μL of assay buffer (agonist assay) or 0.6 μM motilin (0.1 μM final concentration) diluted in assay buffer (antagonist assay)

 $100\,\mu\mathrm{L}$  of [ $^{35}\mathrm{S}$ ]-GTP $\gamma\mathrm{S}$  diluted in assay buffer Final Test Concentrations (N=12) for Compounds: 50, 20, 10, 5, 2, 1, 0.5, 0.2, 0.1, 0.05, 0.02, 0.01  $\mu\mathrm{M}$ . Compound Handling:

Compounds were provided frozen on dry ice at a stock concentration of  $10\,\mathrm{mM}$  diluted in  $100\%\,\mathrm{DMSO}$  and stored at  $-20^\circ$  C. until the day of testing. On the test day, compounds were allowed to thaw at room temperature and than diluted in assay buffer according to the desired test concentrations. Under these conditions, the maximum final DMSO concentration in the assay was 0.5%.

Assay Protocol:

CHO membranes were immobilized into 96-well Flash-Plate® microplates. Test compound, GTPγS, motilin and [35S]-GTPγS were combined in each well according to the Assay Volumes described above.

For the assay to measure agonist activity, an additional 25  $\mu$ l of buffer was added to each well in addition to 25  $\mu$ L of either buffer (basal value, N=4), 1  $\mu$ M (final conc.) motilin (E<sub>max</sub> value, N=3), 25  $\mu$ M (final conc.) GTP $\gamma$ S (non-specific value, N=4), or the appropriate concentration of test compound (N=3).

For the assay to measure antagonist activity, an additional 25  $\mu$ L of either buffer (unstimulated control) or motilin (0.1  $\mu$ M final conc.) is added to each well, in addition to either 25  $\mu$ L of buffer (basal value, N=3), 1  $\mu$ M (final conc.) motilin 15 (E<sub>max</sub> value, N=3), 25  $\mu$ M (final conc.) GTP $\gamma$ S (non-specific value, N=4), or the appropriate concentration of test compound (N=3).

The reaction is initiated by addition of 100 mL of [ $^{35}$ S]-GTP $\gamma$ S to each well. Each plate is sealed (TopSeal-A<sup>TM</sup>) and 20 incubated in the dark at room temperature for 150 min. Then, plates are counted for 30 seconds per well on the TopCount® NYT

Data were analyzed by GraphPad<sup>TM</sup> Prism® 3.0 (GraphPad Software, San Diego, Calif.) using non-linear regression 25 analysis (sigmoidal dose-response) for the calculation of  $IC_{50}/EC_{50}$  values.

$$E_{max}$$
(agonist) or  $D_{max}$ (antagonist) =  $\frac{\text{Top} - \text{Bottom}}{\text{Bottom}} \times 100$ 

Where Top and Bottom correspond to the top and bottom values of the dose-response curve calculated by  $GraphPad^{TM}$  Prism.

#### Example Method B4

# Rabbit Duodenum Contractility Assay

Duodenal segments were vertically suspended in organ chambers of  $10\,\mathrm{mL}$  filled with Krebs buffer and connected to an isotonic force transducer, with a preload of  $1\,\mathrm{g}$ . After a stabilization period, the muscle strips were challenged with  $10^{-4}\,\mathrm{M}$  acetylcholine and washed. This was repeated until a stable maximal contraction was obtained (2-3 times), with an interval of at least  $20\,\mathrm{minutes}$ .

After a stable base line was reached, test compounds were added to the bath. After 15 min incubation, a dose response to motilin was recorded by adding logarithmically increasing concentrations of motilin to the bath (final concentration  $10^{-9}$  to  $10^{-6}$  M). A blank experiment (no test compound present) was also performed. At the end of the dose response curve, a supramaximal dose of acetylcholine ( $10^{-4}$  M) was given and this response was used as a reference (100% contraction).

The results of experiments at different concentrations of test compound were combined and analyzed to derive the  $pA_2$  value from the Schild plot.

It is appreciated that although specific experimental methods have been described herein for the purposes of illustration, various modifications to these experimental methods as well as alternate methods of experimentation may be used without departing from the scope of this invention.

TABLE 3

		Binding activ	rity of selected compounds		
	$R_1$	$R_3$	$R_6$	T	$K_i^{1,2}$
1	OH OH			$\sum_{0}^{Z_3} x$	В
2	when the second		- And	$\sum_{0}^{Z_3}$	A
3	www.			$\sum_{0}^{Z3}$	В
4	The transfer of the transfer o			$\sum_{0}^{Z_3}$	A
5	HO	СН3	- Arrange (	$\sum_{0}^{Z3}$	В

TABLE 3-continued

		Binding activ	vity of selected compounds		
	$R_1$	$R_3$	$R_6$	Т	$K_i^{1,2}$
6	Varyeng OH		No.	Z3 X	В
7	Varyeng OH		- Section of the sect	$\sum_{0}^{Z3}$	В
8	W. C.		www.	$\sum_{0}^{Z3}$	В
9	What OH	-		$Z_3$	В
10	OH	-		$Z_3$	A
11			- Arrivar	$Z_3$	A
12	HO—OH		- m	$Z_3$	В
13	www.		- And	$\sum_{0}^{z_3} x$	В
14			- Sand	$\sum_{0}^{Z_3} x$	В
15	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~			$\sum_{0}^{Z_3} X$	A

TABLE 3-continued

		Binding activ	ity of selected compounds		
	$R_1$	$R_3$	R <sub>6</sub>	T	$K_i^{1,2}$
16	LI CI			$\sum_{0}^{Z_3} x$	A
17	N.H.			X	В
18	Chry OH			X	В
19	The state of the s			X	A
20	wyw \			X	В
21				$\sum_{0}^{Z3}$	A
22				X	A
23		- Andrew Commercial Co	- Section of the sect	$\sum_{0}^{Z_3} X$	A
24	ZZZZZ S	-		$Z_3$	A
25	"CI			$Z_3$	В

TABLE 3-continued

		Binding activity	of selected compounds		
	$R_1$	$R_3$	R <sub>6</sub>	Т	$K_i^{1,2}$
26	No.	-		$Z_3$	A
27	N N N N N N N N N N N N N N N N N N N			$\sum_{0}^{Z3}$	В
28	N. N.	-		$Z_3$	В
29	Very OH	-		X	В
30	Variation OH	-		$\sum_{0}^{Z3}$	В
34	Chryster CH C	OH		$Z_3$	В
38	Varyang OH		$\mathrm{CH_3}$	$\sum_{0}^{Z_3}$	С
39	The Contract of the Contract o		Н	$Z_3$	В
40	Variation OH		Н	$Z_3$	С
41	why			$\sum_{0}^{Z_3} X$	С
52	Contrar OH	-		$Z_3$	В

TABLE 3-continued

		Binding act	ivity of selected compounds		
	$R_1$	$R_3$	$R_6$	Т	$K_i^{1,2}$
55	OH OH		NH <sub>2</sub>	Z3 X	В
56	Wyky, OH		NH <sub>2</sub>	$\sum_{0}^{Z3}$	В
57	Wyky, OH		OH	$\sum_{0}^{Z_3} x$	В
58	HO		OH	$\sum_{0}^{Z_3} x$	В
59	OH		NH NH	$Z_3$ $X$	В
60	OH		~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	X	С
65	OH			x Z3	В
71				O Z3	В
72	OH OH	-		X	В
76	What of the second of the seco	nym S		$Z_3$	С

TABLE 3-continued

Binding activity of selected compounds					
$R_1$	$R_3$	$R_6$	T	$K_i^{1,2}$	
77 OH	- Amount		Z3 X	С	
OH OH			$Z_3$	В	
85 OH		Н	$Z_3$	В	
87		- www.	$Z_3$	В	
0 O			$Z_3$	С	
89			$\sum_{0}^{Z3}$	С	
90		- Sampana	$Z_3$	С	
91 OH			$\sum_{0}^{Z_3} x$	С	
92 OH	- And -		$\sum_{0}^{Z3}$	В	
OH OH	Н		X	С	

TABLE 3-continued

Binding activity of selected compounds					
$R_1$	$R_3$	$R_6$	Т	$K_i^{1,2}$	
97 North OH	2240	- Announce of the contract of	$Z_3$	С	
HO————————————————————————————————————			$X \longrightarrow \stackrel{H}{N} \longrightarrow_{Z3}$	С	
OH OH			$X$ $N$ $Z_3$	С	
109 ANALO OH		NH <sub>2</sub>	$Z_3$	В	
OH OH			X	В	
OH OH			$Z_3$	В	
112 OH	-		$Z_3$	В	
122 N	-		$Z_3$	В	
123 N			$Z_3$	В	
124 N	-		$\sum_{0}^{Z_3} X$	В	
125 ОН		- Sacration	$Z_3$	В	

TABLE 3-continued

Binding activity of selected compounds					
	$R_1$	$R_3$	$R_6$	T	$K_i^{1,2}$
126	ОН			$Z_3$	В
127	H N N			X Z3 S	В
128				X	В
129				$Z_3$	A
130	s ·····			$\sum_{0}^{Z_3} X$	В
131	CI			$\sum_{0}^{Z_3} X$	A
132	CI			X	A
133	F	- Marie 1		X	A
134	CI			CI	С
135		- Market   M		$\sum_{i=1}^{N} z_{i}$	В

TABLE 3-continued

Binding activity of selected compounds					
$R_1$	$R_3$	$R_6$	Т	$K_i^{1,2}$	
136			X	В	
137 S	-		X	В	
138 CI			X	В	
139			X	В	
140 F			X	В	
141			$Z^3$	A	
142			$Z_3$	A	
143 S			$\sum_{0}^{Z3}$	В	
144	- www.		$Z^3$	A	

TABLE 3-continued

		Binding activ	rity of selected compounds		
	$R_1$	$R_3$	$R_6$	T	$K_i^{1,2}$
146	CI			Z3 X	A
147	CI			X	В
148	F	- Marie 1		$ \begin{array}{c c}  & Z3 \\ \hline  & X \end{array} $	В
149				$\sum_{i=1}^{N} z_{i}$	A
150	CI			X	В
151	F			$ \begin{array}{c c}  & Z3 \\  & X \end{array} $	В
152		- Marie - Mari	NH <sub>2</sub>	$Z_3$	В
153			NH <sub>2</sub>	$Z_3$	В
154	s		NH <sub>2</sub>	$Z_3$	В
155	CI	- <del>**</del>	NH <sub>2</sub>	$\sum_{0}^{z_3}$	A

TABLE 3-continued

	Binding activity of selected compounds					
	$R_1$	$R_3$	$R_6$	T	$K_i^{1,2}$	
156	CI		NH <sub>2</sub>	$\sum_{0}^{Z_3} x$	A	
157			NH <sub>2</sub>	$Z_3$	В	
158	CI	-		X	A	
159	OH			X	В	
160	OH			X	В	
161	NH <sub>2</sub>			$Z_3$	В	
162			- Sand	$\sum_{0}^{Z_3} X$	В	
163				$\sum_{0}^{Z_3} x$	A	
164		-		$\sum_{0}^{Z_3} x$	В	
165				X	A	
166				X	В	

TABLE 3-continued

	Binding activ	ity of selected compounds		
$R_1$	$R_3$	$R_6$	T	$K_i^{1,2}$
167 S	- Andrews		$Z_3$ $X$	A
168 S	- \$ - <		$Z_3$	A
169 Cl			$Z_3$	В
170 Cl			$\sum_{0}^{Z_3}$	A
171 F	-		$Z_3$	A
172	*		$Z_3$	A
173			$\sum_{0}^{Z3}$	В
174			$Z_3$	В
175	- Marie 1		$\sum_{0}$ $\sum_{X}$	В
176	- Marie 1		$Z_3$	В
177			$Z_3$	В

TABLE 3-continued

		Binding acti	ivity of selected compounds		
	$R_1$	$R_3$	R <sub>6</sub>	Т	$K_i^{1,2}$
178		-	ОН	$\sum_{0}^{Z_3} X$	В
179			ОН	$\sum_{0}^{Z_3} X$	В
180	s · ··································		ОН	$\sum_{0}^{Z_3}$	В
181	Cl	-	ОН	$\sum_{0}^{Z_3} X$	A
182	CI		ОН	$\sum_{0}^{Z_3} X$	A
183	F		ОН	$Z_3$	В
184			NH <sub>2</sub>	X	В
185	CI			X	В
186	ОН		NH	$\sum_{0}^{23} x$	В
187	F			$\sum_{0}^{Z3}$	Α

TABLE 3-continued

Binding activity of selected compounds					
	$R_1$	$R_3$	$R_6$	T	$K_i^{1,2}$
188	F			$\sum_{0}^{Z3} X$	A
189				$\sum_{0}^{Z3} X$	В
190	S S			$Z_3$ $X$	A
191				$\sum_{0}^{Z3} x$	A
192	CI			$\sum_{0}^{Z_3} X$	A
193	F			$\sum_{0}^{Z3} X$	A
194		- www.		$Z_3$	В
195	• 0			$Z_3$	A
196	CI			$H_2N$ $X$	

TABLE 3-continued

			ivity of selected compounds		
	$R_1$	$R_3$	$R_6$	Т	K <sub>i</sub> <sup>1,2</sup>
197	CI			O $X$ $X$	
198	CI	-		$Z_3$	A
199	CI	-		Z3 $HO$ $OH$ $X$	В
200	CI	-		$Z_3$	A
201				$Z_3$	В
202				$\sum_{0}^{Z3} X$	A
203	CI		HN	$Z_3$	В
204	CI	-	NH NH <sub>2</sub>	$Z_3$	A
205	CI	- See - Company		$Z_3$	В
206	CI		HN — S — O	$z_3$	В

TABLE 3-continued

			ivity of selected compounds		
	$R_1$	$R_3$	$R_6$	T	K <sub>i</sub> <sup>1,2</sup>
207	CI		HN NH <sub>2</sub>	Z3 X	В
208	OH		NH NH <sub>2</sub>	$Z_3$	В
209	OH	-	HN—O	$\sum_{0}^{Z_3} x$	С
210	OH		HN	$\sum_{0}^{Z3}$	
211	CI			$Z_3$	A
212	CI	-	NH	$Z_3$	A
213	OH			$Z_3$	В
214			ОН	$C_0$ $C_2$ $C_3$ $C_2$ $C_3$ $C_4$ $C_4$ $C_5$	В
215	OH	-	N—	$\sum_{0}^{Z_3} X$	В
216	CI	-		$\sum_{0}^{Z3}$	A
217	CI			F $Q$ $X$	В

TABLE 3-continued

			rity of selected compounds		
	$R_1$	$R_3$	$R_6$	Т	$K_i^{1,2}$
218	CI			$Z_3$	A
219	Cl	- Marie 1		Cl $Z3$ $X$	В
220	F			$Z_3$	A
221				F 0 X	В
222	OH			$\sum_{0}^{Z3} X$	A
223	OH			$Z_3$	С
224		- Anna		$Z_3$	В
225		- And		$\sum_{0}^{Z3} X$	В
226				$Z_3$	С

#### TABLE 3-continued

Binding activity of selected compounds					
	$R_1$	$R_3$	$R_6$	T	$K_i^{1,2}$
227		- save		$Z_3$ $X$	В

25

30

45

50

60

65

Notes

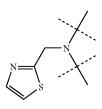
Radioligand competitive binding assays performed using Method B1 Values reported as ranges: A = 0.001-0.100  $\mu M;$  B = 0.100-1.0  $\mu M;$  C = 1.0-10.0  $\mu M$ 

# X is NH except for:

Compound 223 and 225, X is:



Compound 224, X is NMe Compound 226, X is:



Compound 227, X is

 $Z_1,Z_2$  and  $Z_3$  are NH except for compounds 30, 173 and 174  $\,^{55}$  and where  $Z_1$  is O and compound III where  $Z_2$  is O.

 $R_2$ ,  $R_4$  and  $R_5$  are hydrogen except for compound 85 where it

What is claimed is:

1. A compound of formula W-T-Y, wherein W and Y are independently selected from the group consisting of —OH, —NH $_2$  and —NHR $_{60}$ , wherein R $_{60}$  is lower alkyl, and T is the following structure:

$$\begin{array}{c|c} & & \\ & &$$

ss wherein

 $L_5$  is CH or N;

L<sub>6</sub> is CR<sub>52</sub>R<sub>53</sub> or O, wherein R<sub>52</sub> and R<sub>53</sub> are independently selected from the group consisting of hydrogen and lower alkyl;

0  $R_{54}$  and  $R_{55}$  are independently lower alkyl; and, when  $L_6$  is  $CR_{52}R_{53}$ ,  $R_{55}$  is also selected from hydroxy;

 $R_{56}$  is hydrogen or lower alkyl, and  $R_{57}$  is selected from the group consisting of hydrogen and lower alkyl, and, when no double bond is present between the carbon atoms to which  $R_{56}$  and  $R_{57}$  are bonded,  $R_{56}$  and  $R_{57}$  are also independently selected from hydroxy; and

(W) indicates the point of attachment of T to W; and (Y) indicates the point of attachment of T to Y;

with the proviso that T is not the following:

$$(Y)$$

$$(W),$$

$$(Y)$$

$$(Y)$$

$$(Y)$$

$$(Y),$$

$$(Y),$$

$$(Y),$$

m,  $n_1$  and p are zero.

further comprising one or more protecting groups.

10

15

2. A compound of formula W-T-Y, wherein W and Y are independently selected from the group consisting of —OH, —NH $_2$  and —NHR $_{60}$ , wherein R $_{60}$  is lower alkyl, and T is the following structure:

$$R_{56}$$
 $R_{57}$ 
 $R_{57}$ 
 $R_{55}$ 
 $R_{54}$ 

wherein

L<sub>5</sub> is CH or N;

 $L_6$  is  $CR_{52}R_{53}$  or O, wherein  $R_{52}$  and  $R_{53}$  are independently selected from the group consisting of hydrogen and  $^{20}$  lower alkyl;

 $R_{54}$  and  $R_{55}$  is lower alkyl; and, when  $L_6$  is  $CR_{52}R_{53}$ ,  $R_{55}$  is also selected from hydroxy;

 $R_{56}$  and  $R_{57}$  are independently selected from the group consisting of hydrogen and lower alkyl, and, when no double bond is present between the carbon atoms to which  $R_{56}$  and  $R_{57}$  are bonded,  $R_{56}$  and  $R_{57}$  are also independently selected from hydroxy; and

(W) indicates the point of attachment of T to W; and (Y) indicates the point of attachment of T to Y; with the proviso that T is not the following:

3. The compound of claim 2, further comprising one or more protecting groups.

\* \* \* \* \*

# UNITED STATES PATENT AND TRADEMARK OFFICE

# **CERTIFICATE OF CORRECTION**

PATENT NO. : 9,181,298 B2 Page 1 of 1

APPLICATION NO. : 12/273638

DATED : November 10, 2015 INVENTOR(S) : Marsault et al.

It is certified that error appears in the above-identified patent and that said Letters Patent is hereby corrected as shown below:

# In the specification

Column 4, Line 2: Please correct "(=O-O," to read -- (=O)-, --

Column 71, Compounds 143 and 144: Please correct the first compound on each row to read as follows:

Signed and Sealed this Fifteenth Day of November, 2016

Michelle K. Lee

Vichelle K. Lee

Director of the United States Patent and Trademark Office